

Geochemistry and isotopic characterization of volcanic gases for source, degassing, and MTI studies

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Typical High Temp (920°C) volcanic gas from arc volcano (mol %)

Direct gas sampling:

H_2O	95	N_2	0.025
CO_2	1.6	He	0.00014
SO_2	1.3	H_2	0.77
H_2S	0.4	O_2	<0.0005
HCl	0.7	CH_4	0.00005
HF	0.01	CO	0.0008

C, N, S, H, O, noble gas isotopes

... and many trace elements (PGE, Na, K, Sr, Rb, B, Be...) at ppm, ppb levels

Kudryavy Volcano, Kurile Islands: Fischer et al., EPSL 1998



MultiGAS (i.e. Aiuppa)

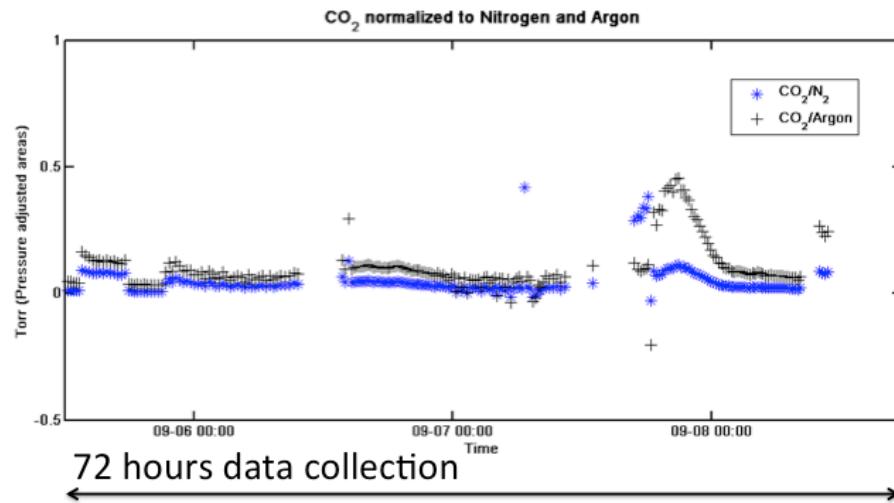
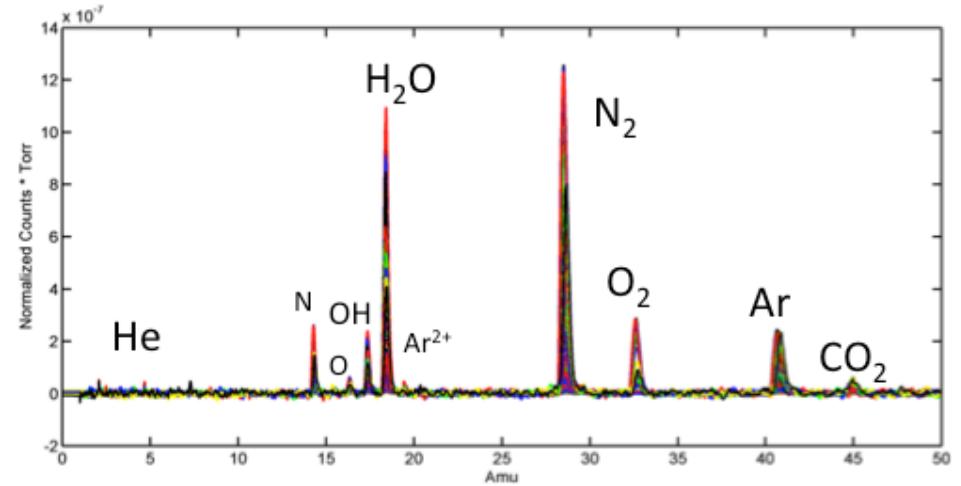
SO_2 , H_2S , CO_2 (H_2O , H_2)



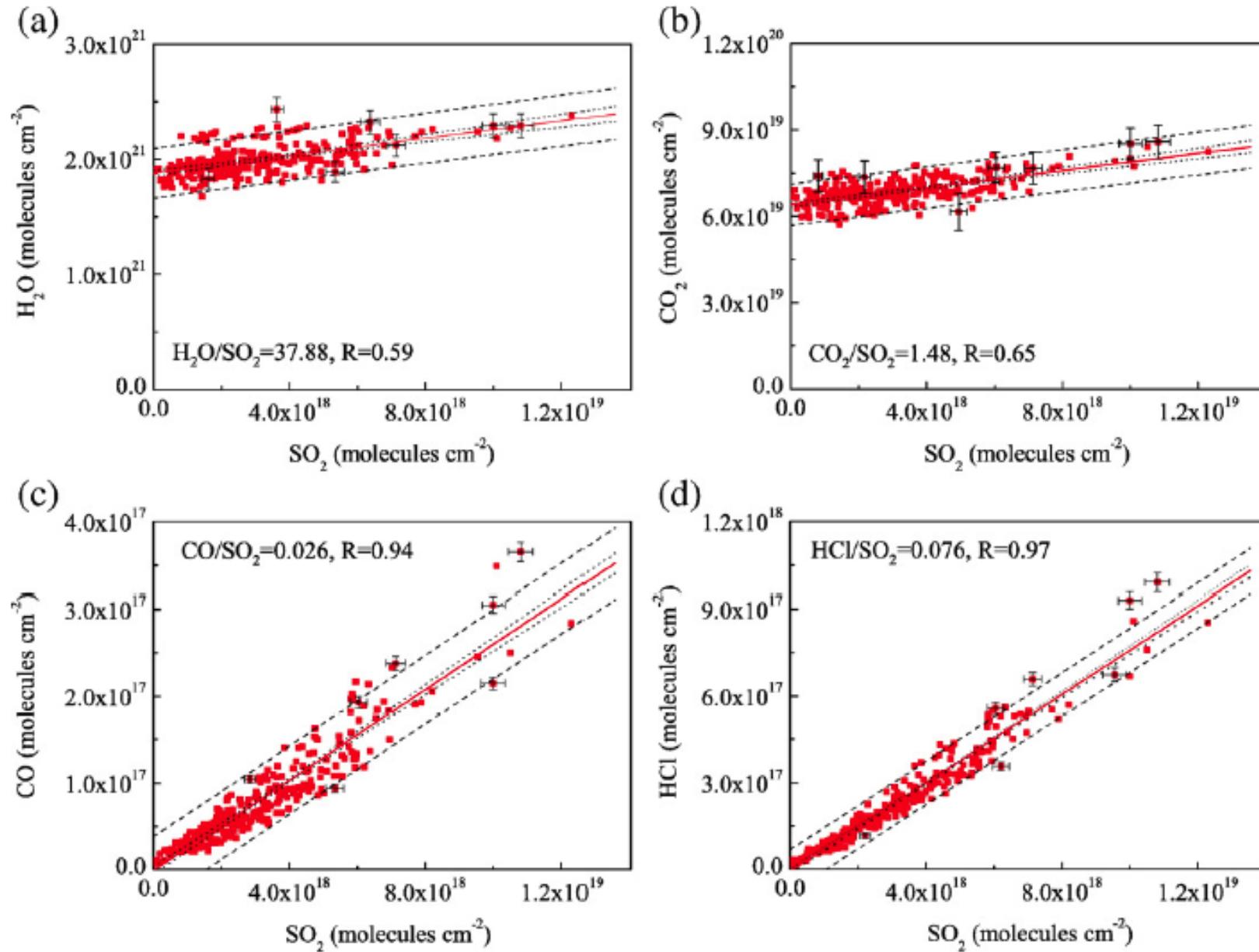
FTIR (i.e. Oppenheimer)

SO_2 , H_2S , CO_2 , CO , HCl

Volcano Mass Spectrometer



multiGAS and FTIR: lots of data, remote sensing/sampling





GIGGENBACH
BOTTLES

SCIENTIFIC GLASSCRAFT

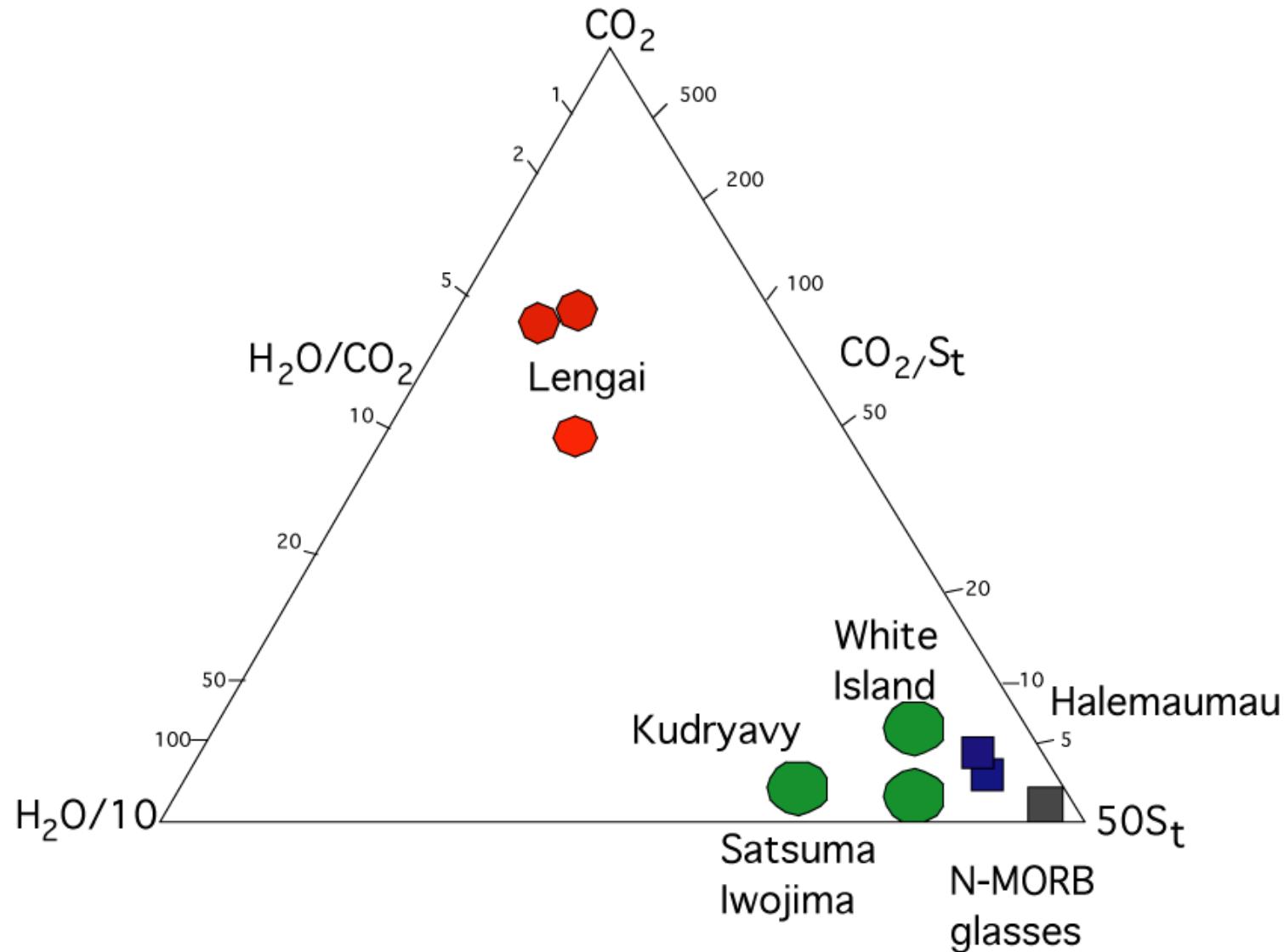
SCIENTIFIC GLASSBOTTLES LTD

NEW ZEALAND

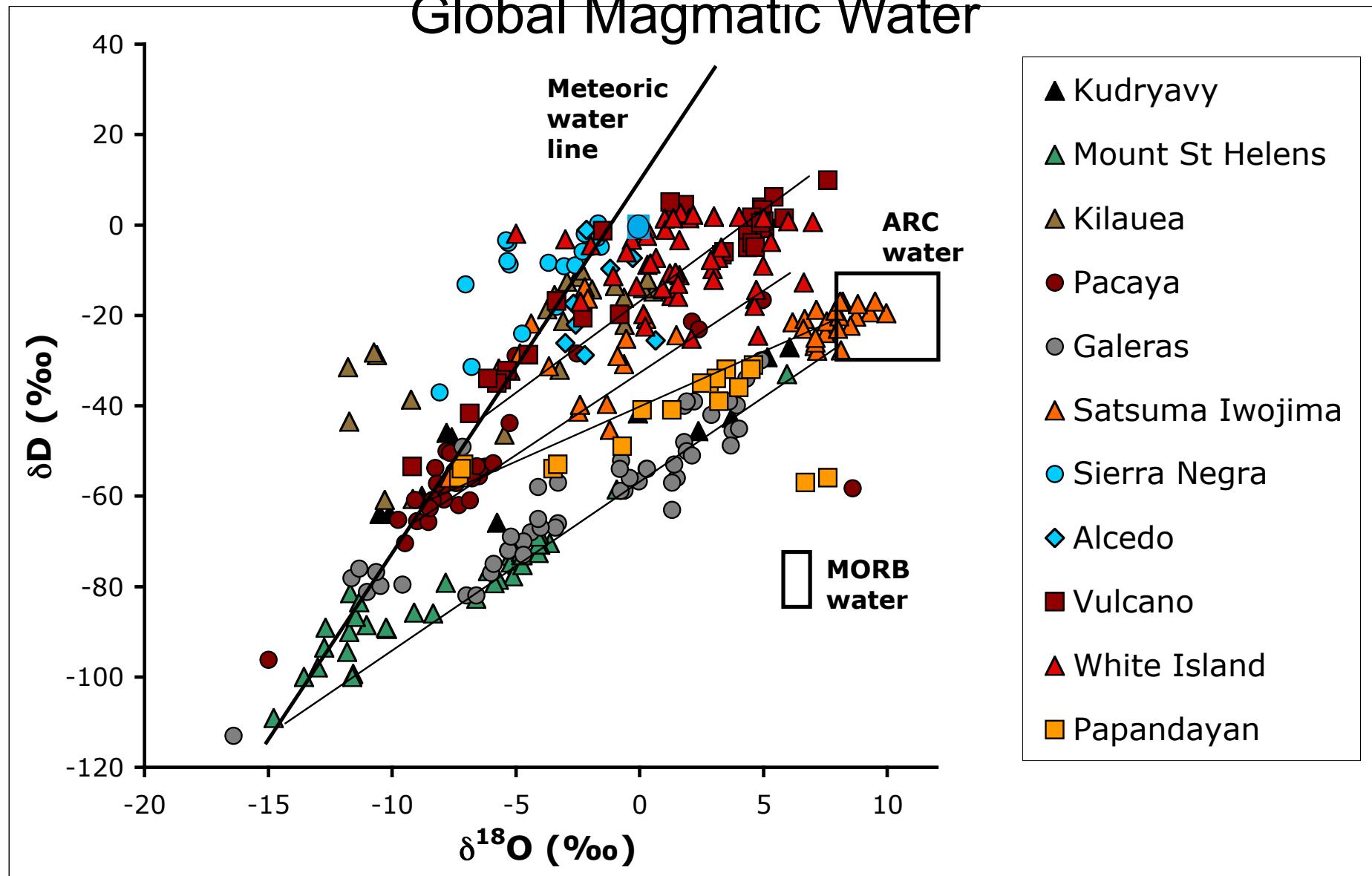
FAX

+ 64 4 586 1591

Global Compositions: gas samples



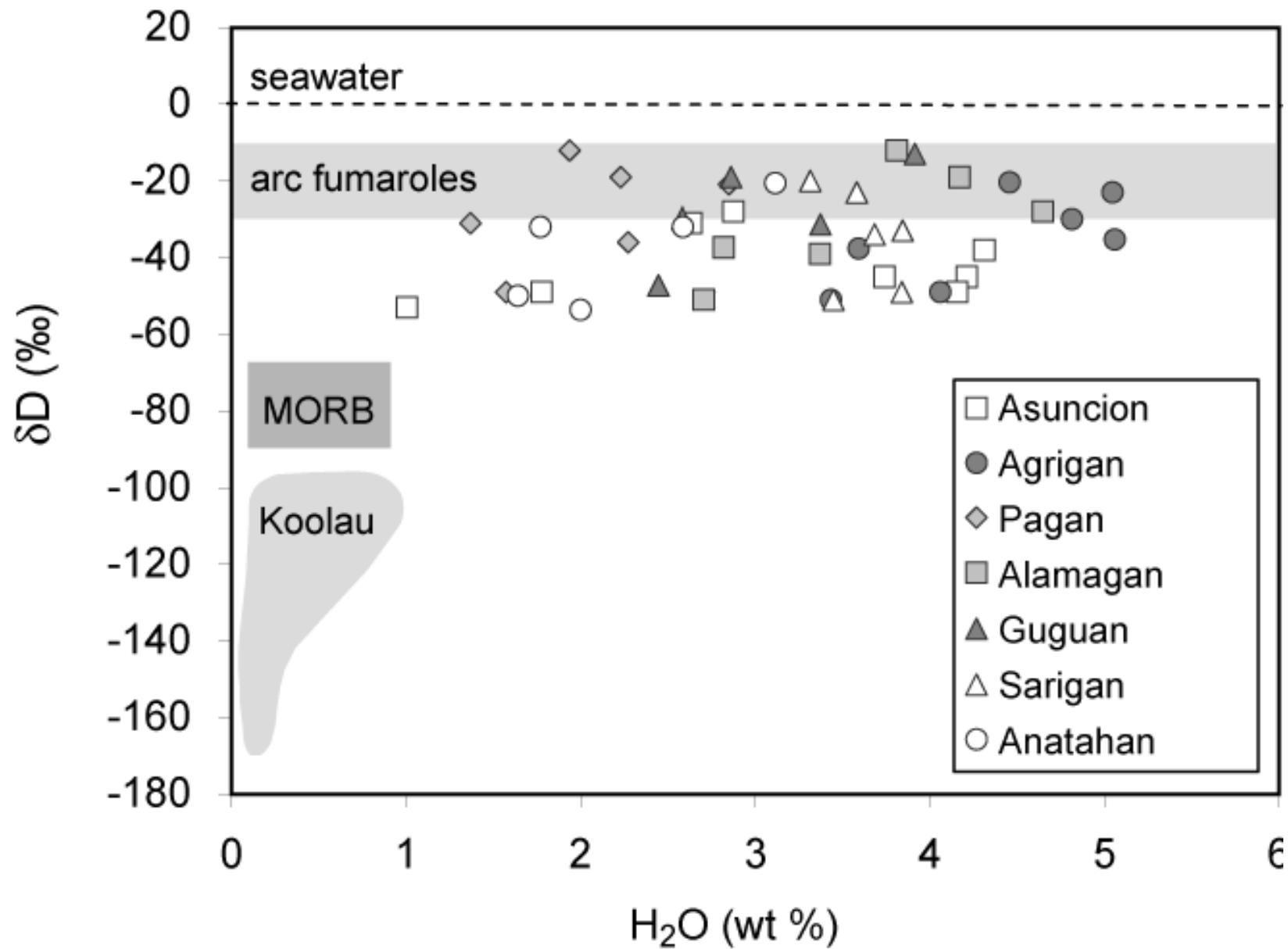
Global Magmatic Water



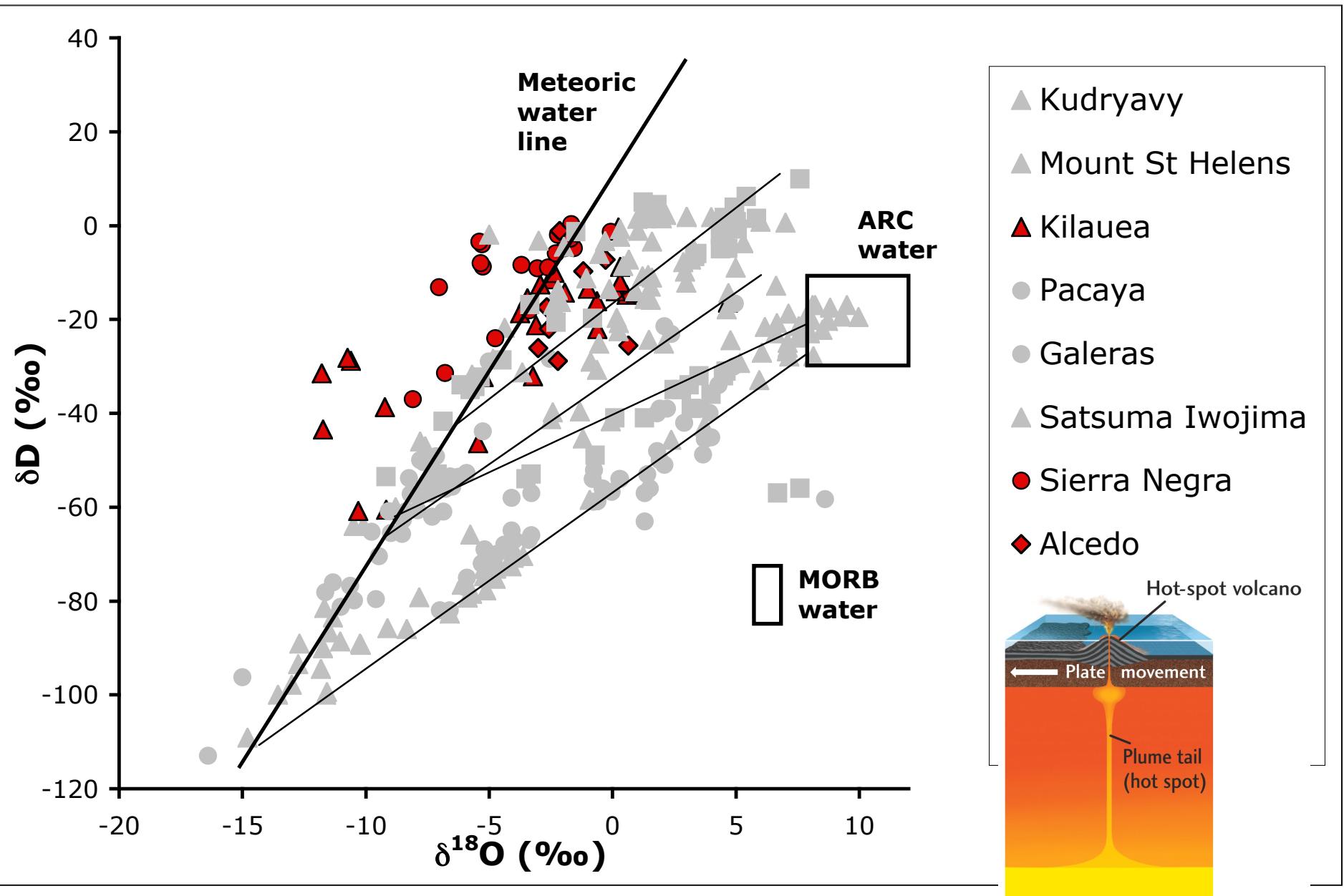
After Giggenbach 1992 and Taran 1992

from Oppenheimer, Fischer, Scaillet (in press)

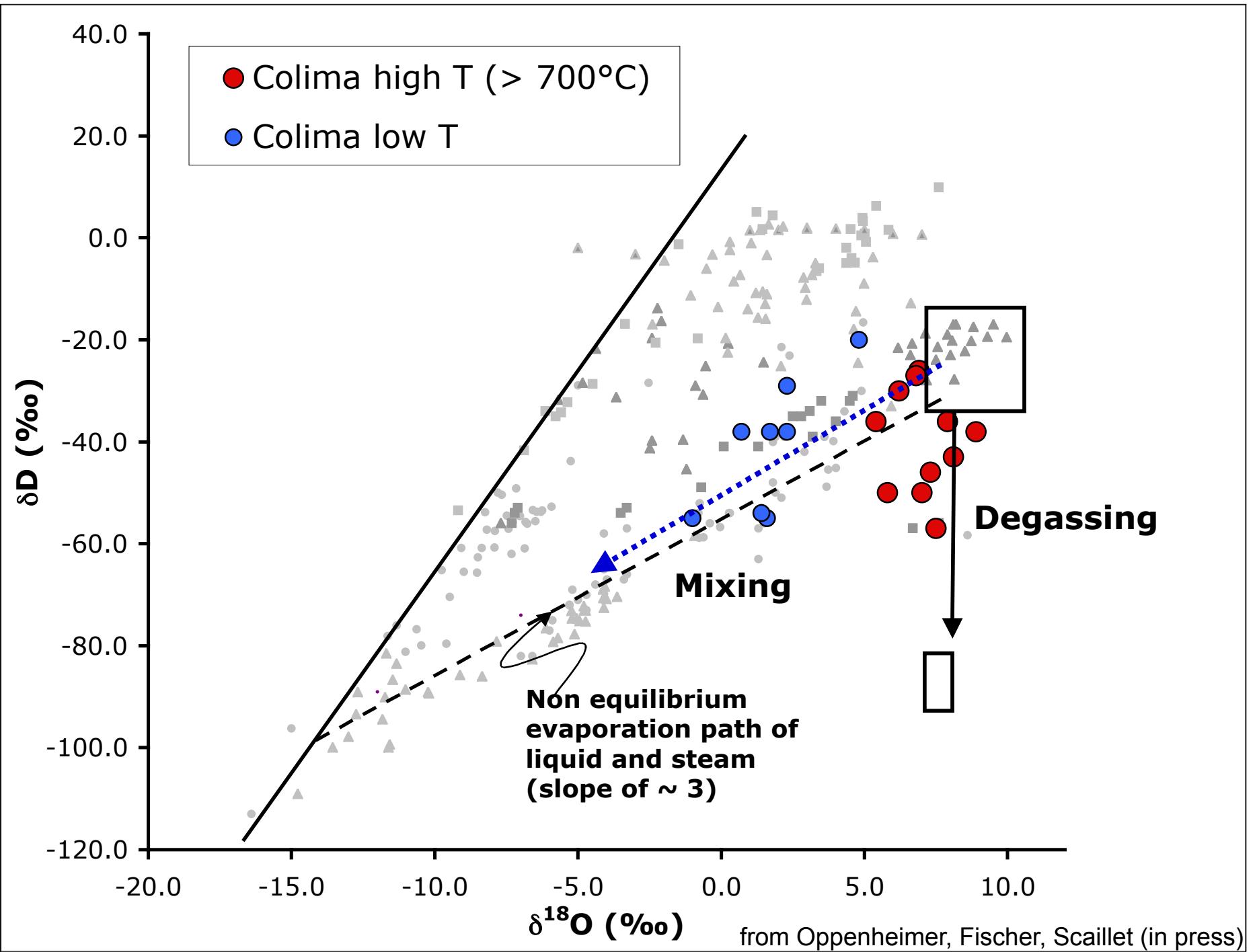
Melt inclusions: Marianas



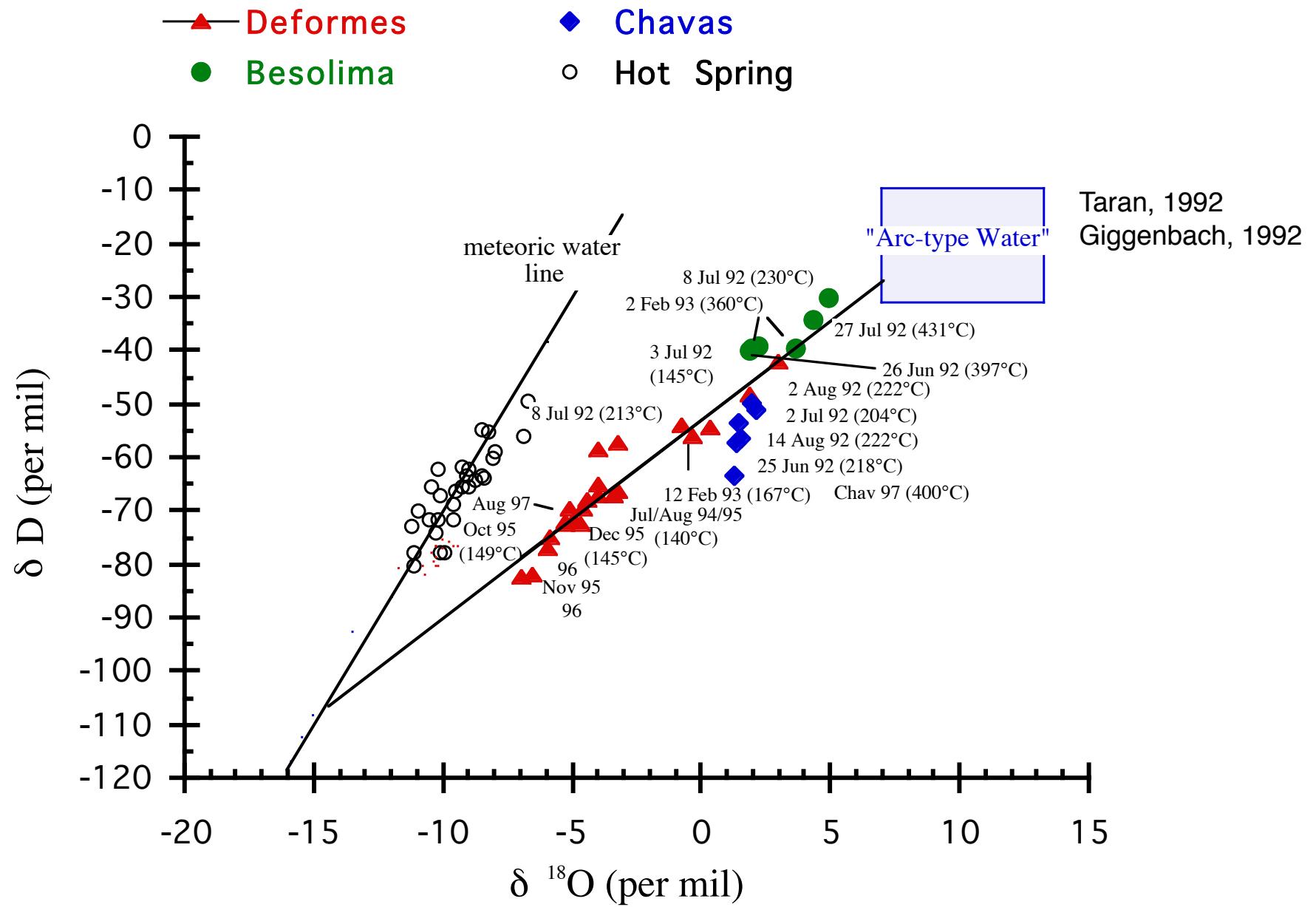
Shaw, Hauri, Fischer et al. EPSL 2008



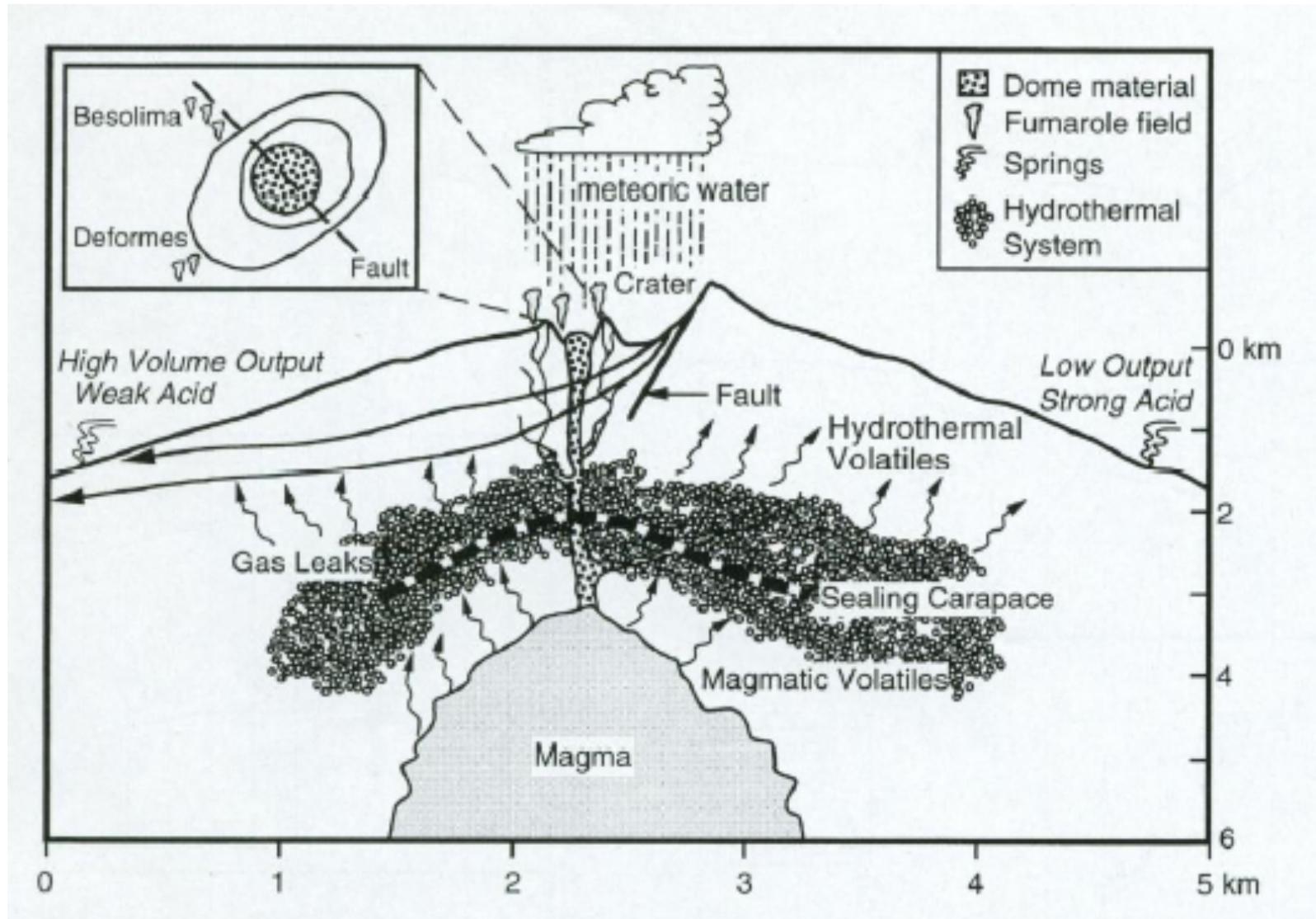
from Oppenheimer, Fischer, Scaillet (in press)



Galeras Volcano, Colombia: Oxygen and Hydrogen isotope variations of steam

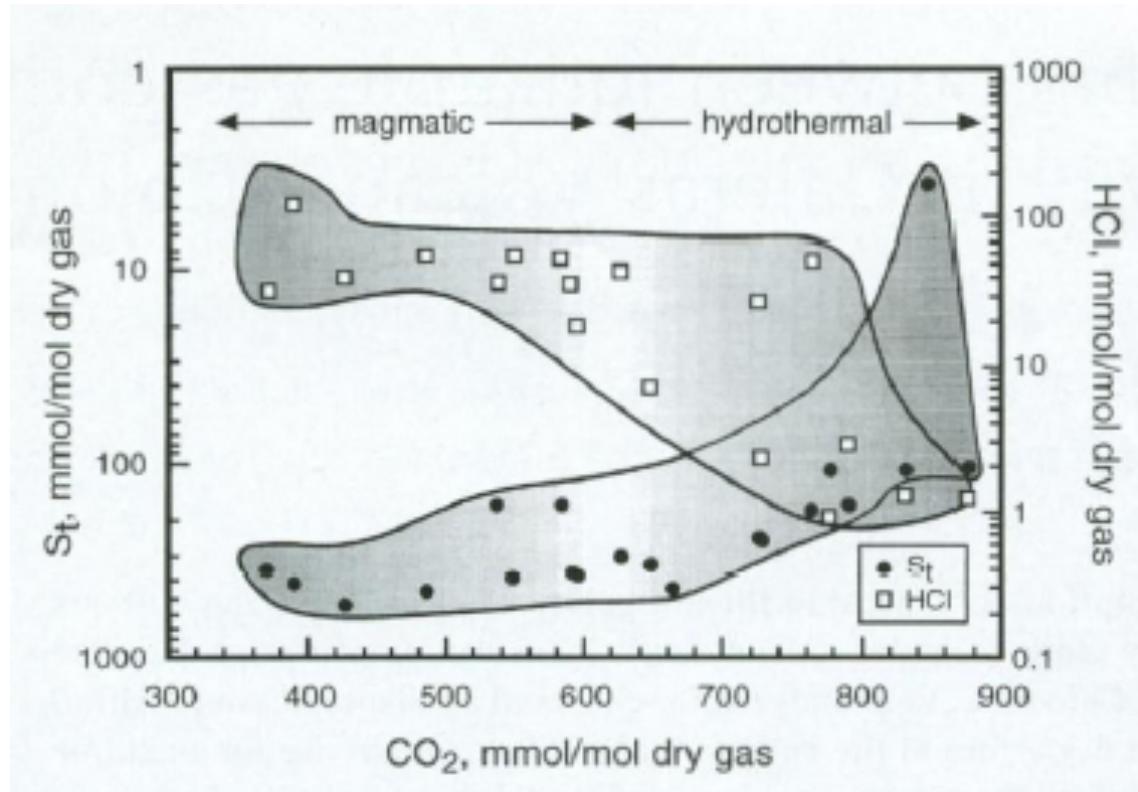


Fischer et al., 1997



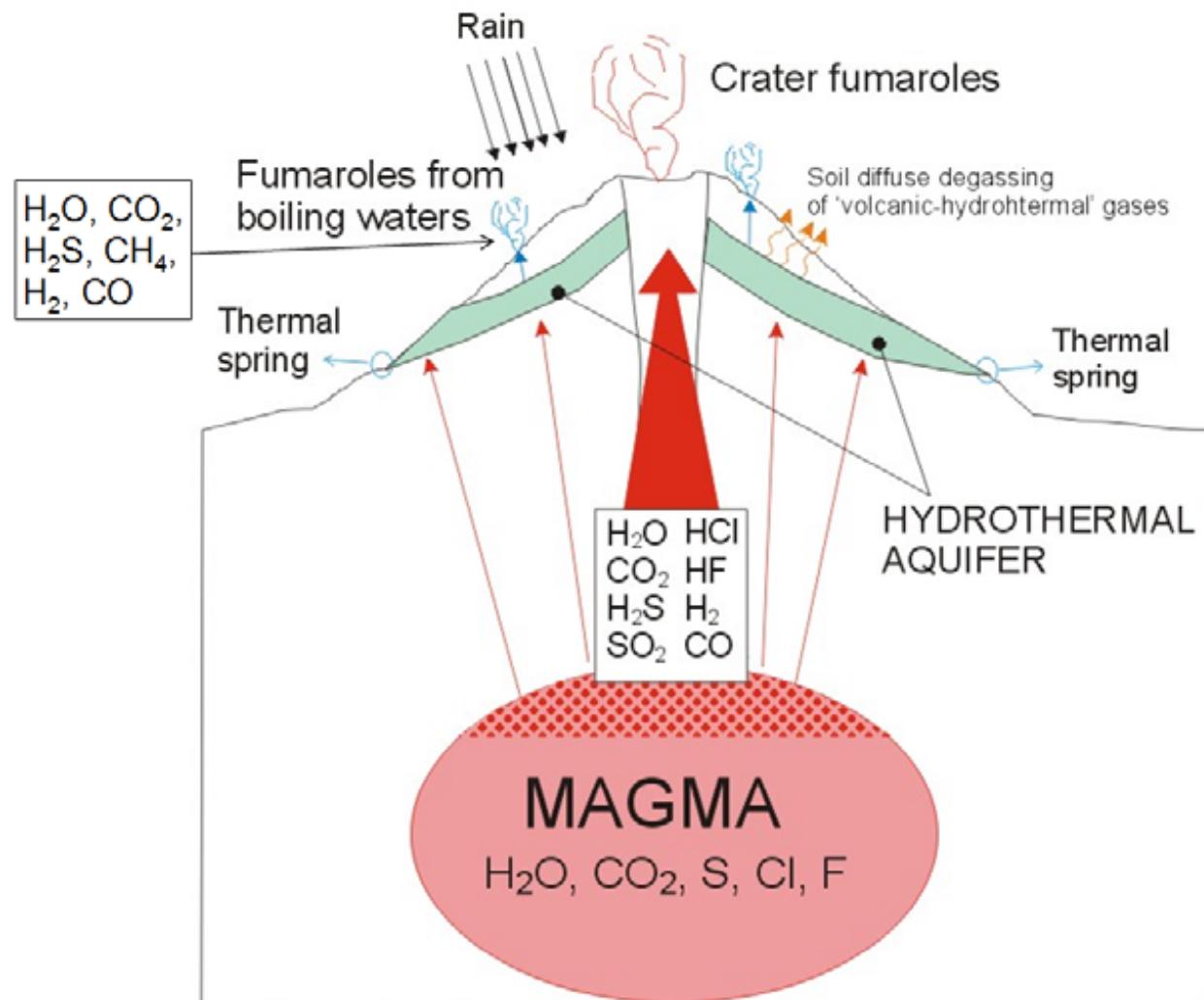
Fischer et al., 1996

Magmatic vs hydrothermal

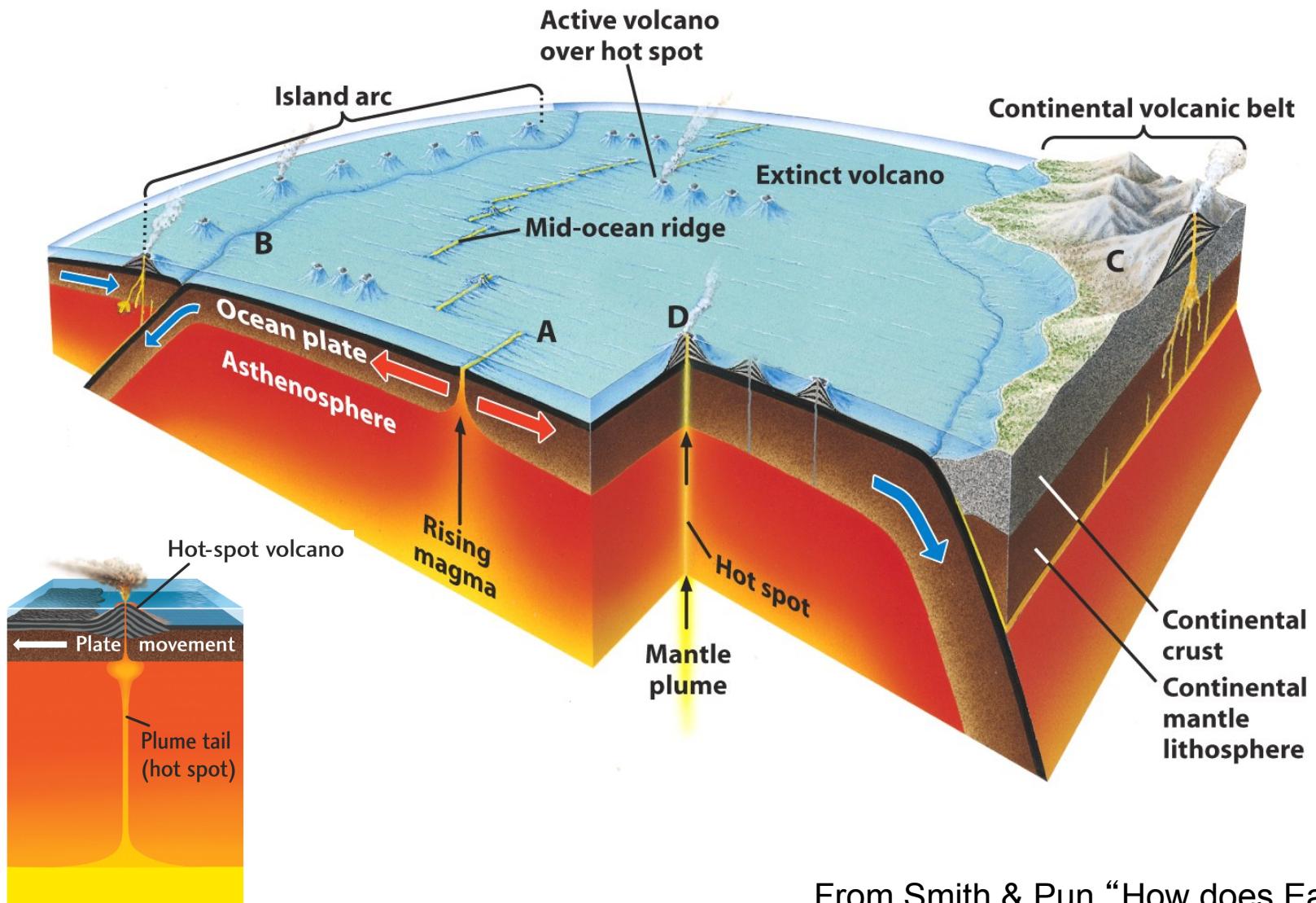


Fischer et al., 1996

Volcano from gas geochemist's perspective (from Chiodini, pers. communication, 2013)



Volatile sources: the bigger picture

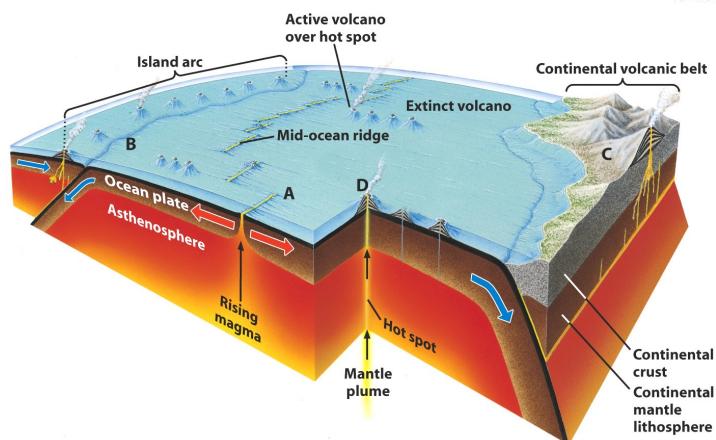
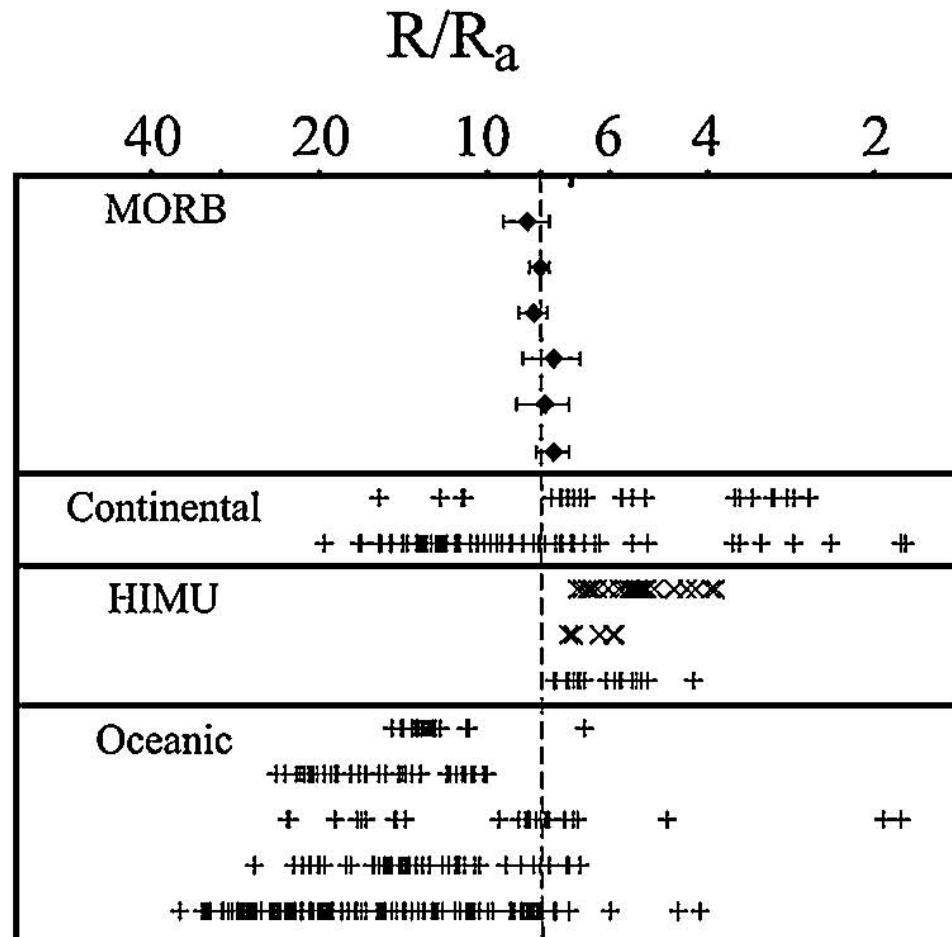


From Smith & Pun "How does Earth Work?"

Helium

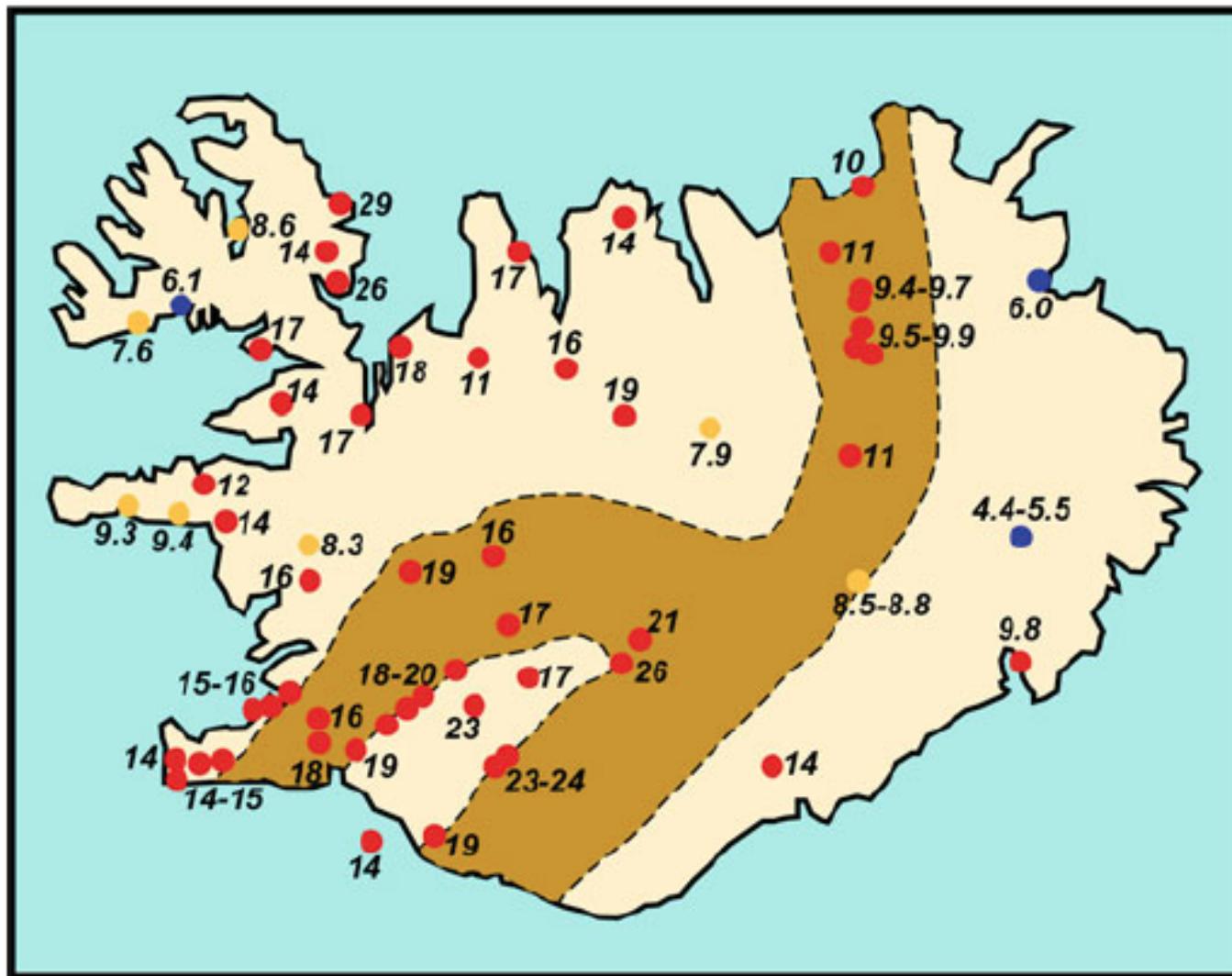
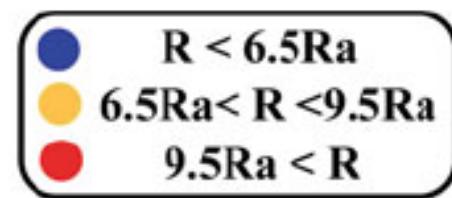
OIBs

- East Pacific Rise
- Juan de Fuca
- Central Indian and Carlsberg Ridges
- South West Indian Ridge
- North Atlantic
- South Atlantic
- Yellowstone & Columbia River Basalt Group
- Afar
- Cameroon line continental
- Cameroon line oceanic
- St Helena, Mangaia and Tubuai
- Reunion
- Samoa
- Galapagos
- Iceland
- Hawaii

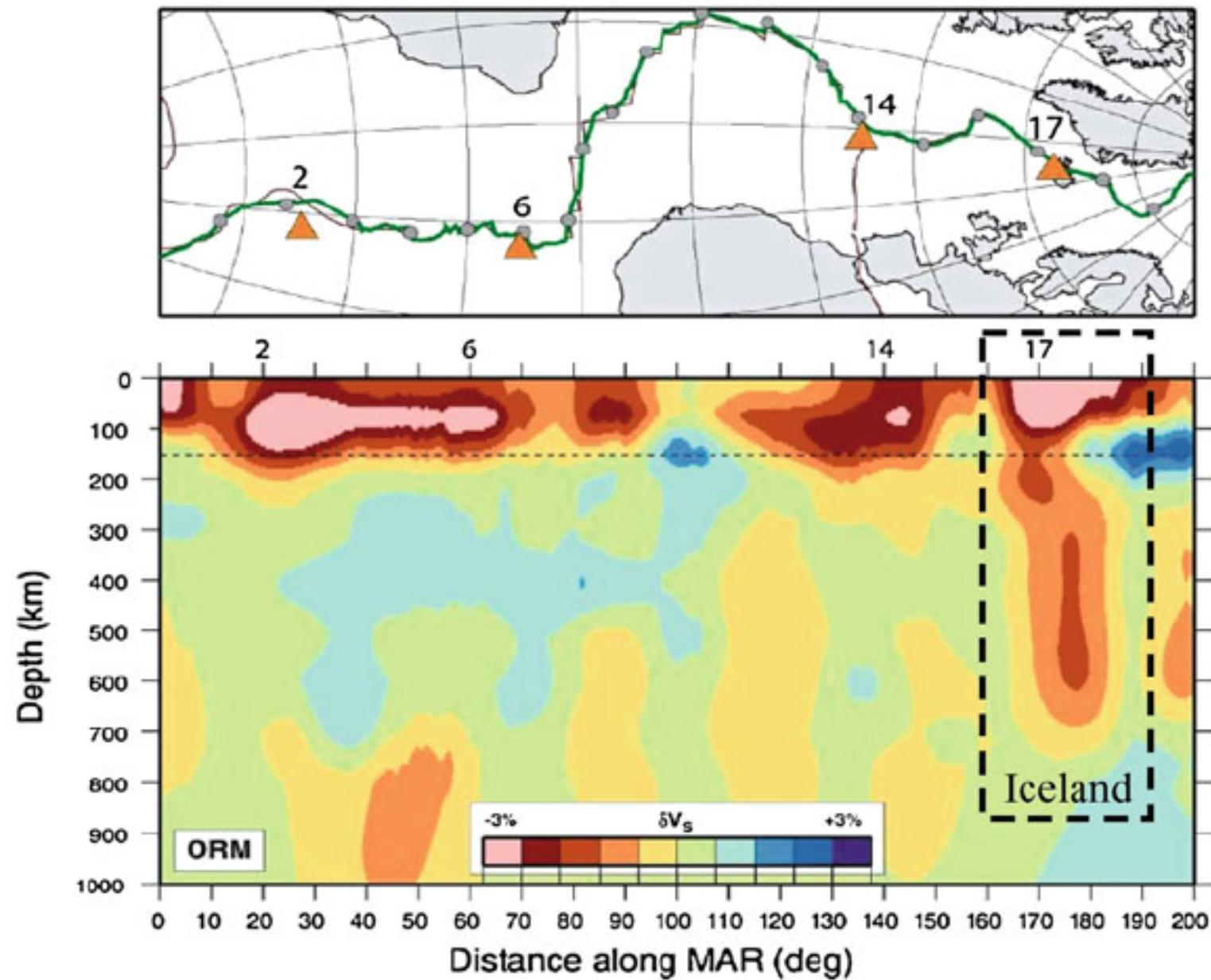


Van Keken *et al.*, 2002

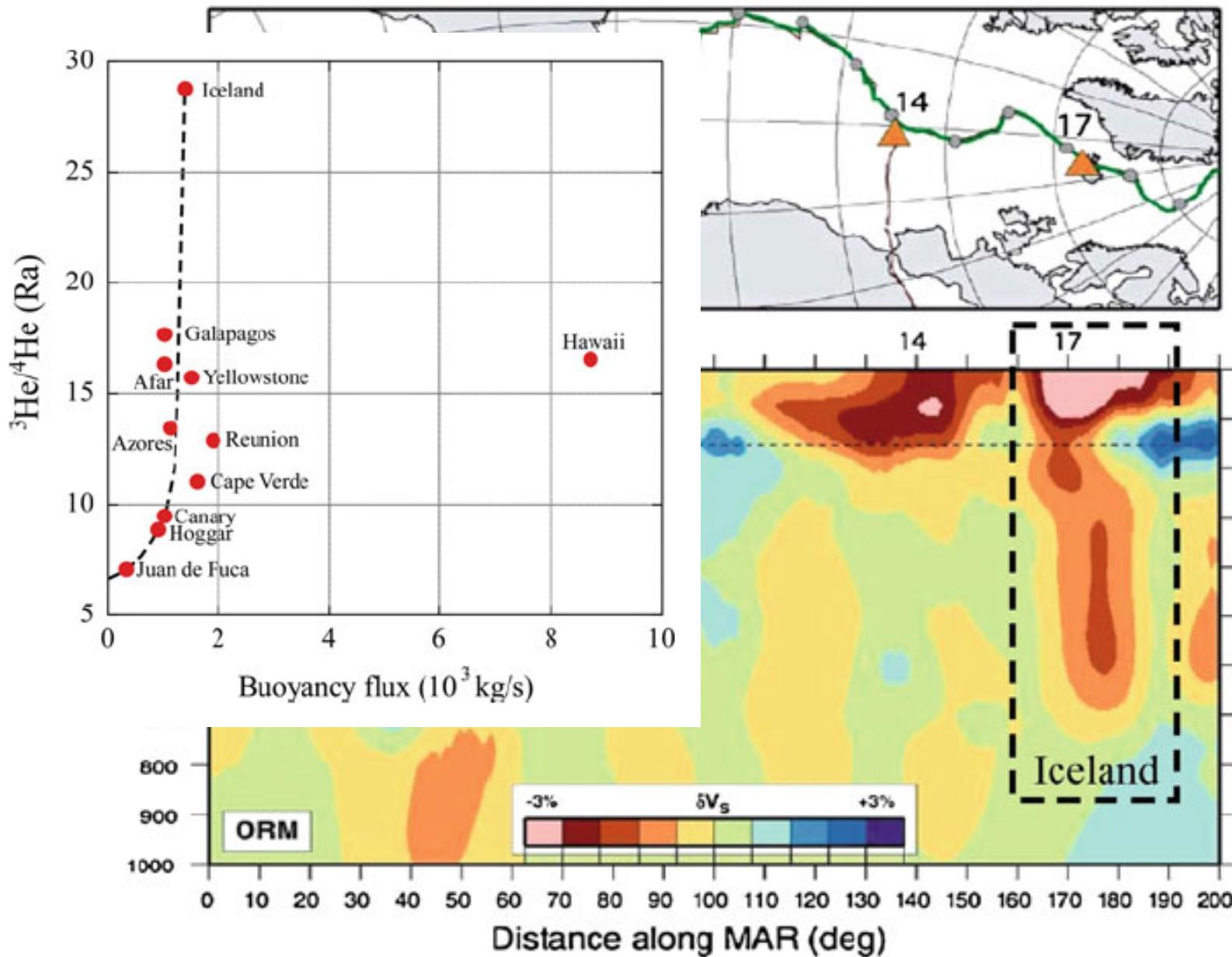
Iceland Fluids



Sano and Fischer, 2013

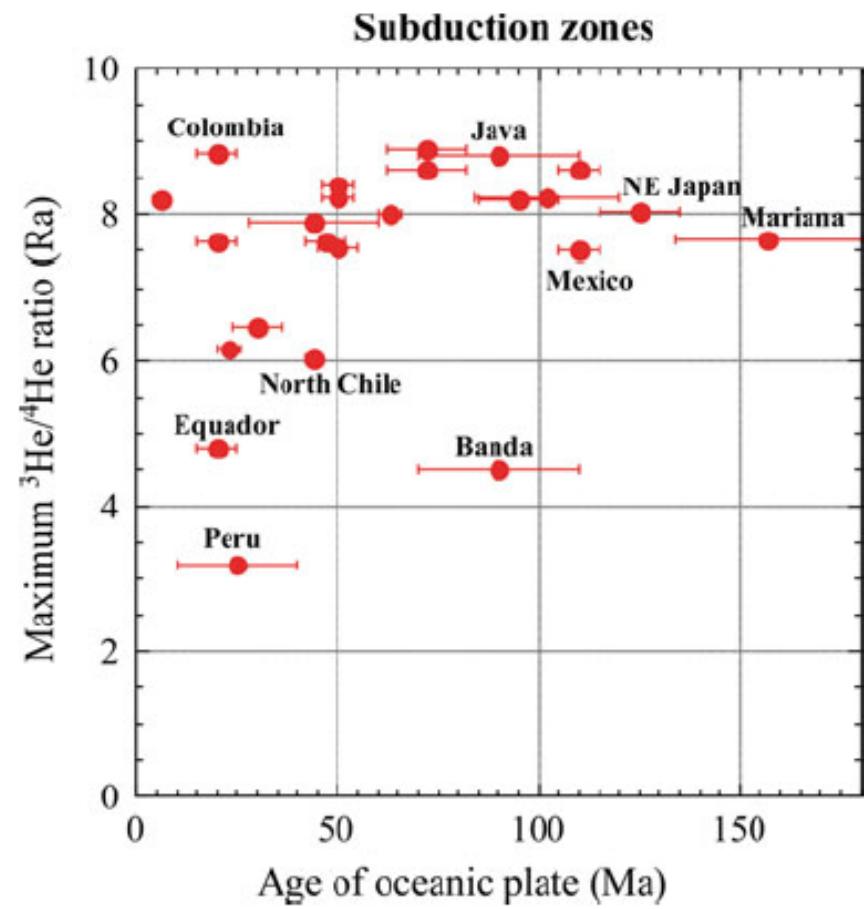
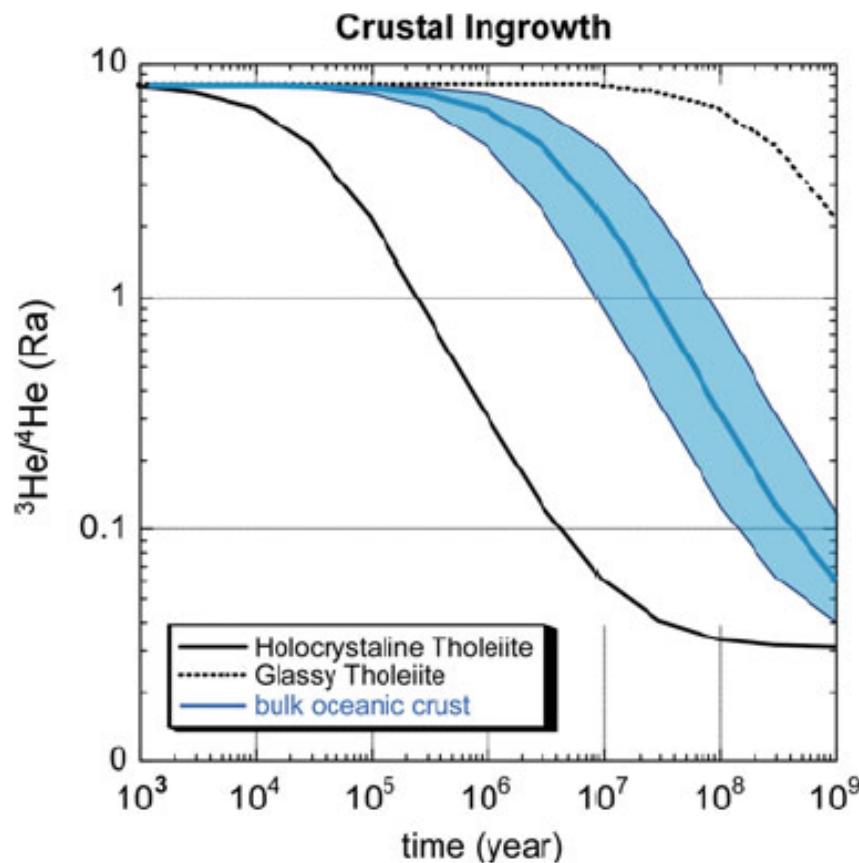


Sano and Fischer, 2013

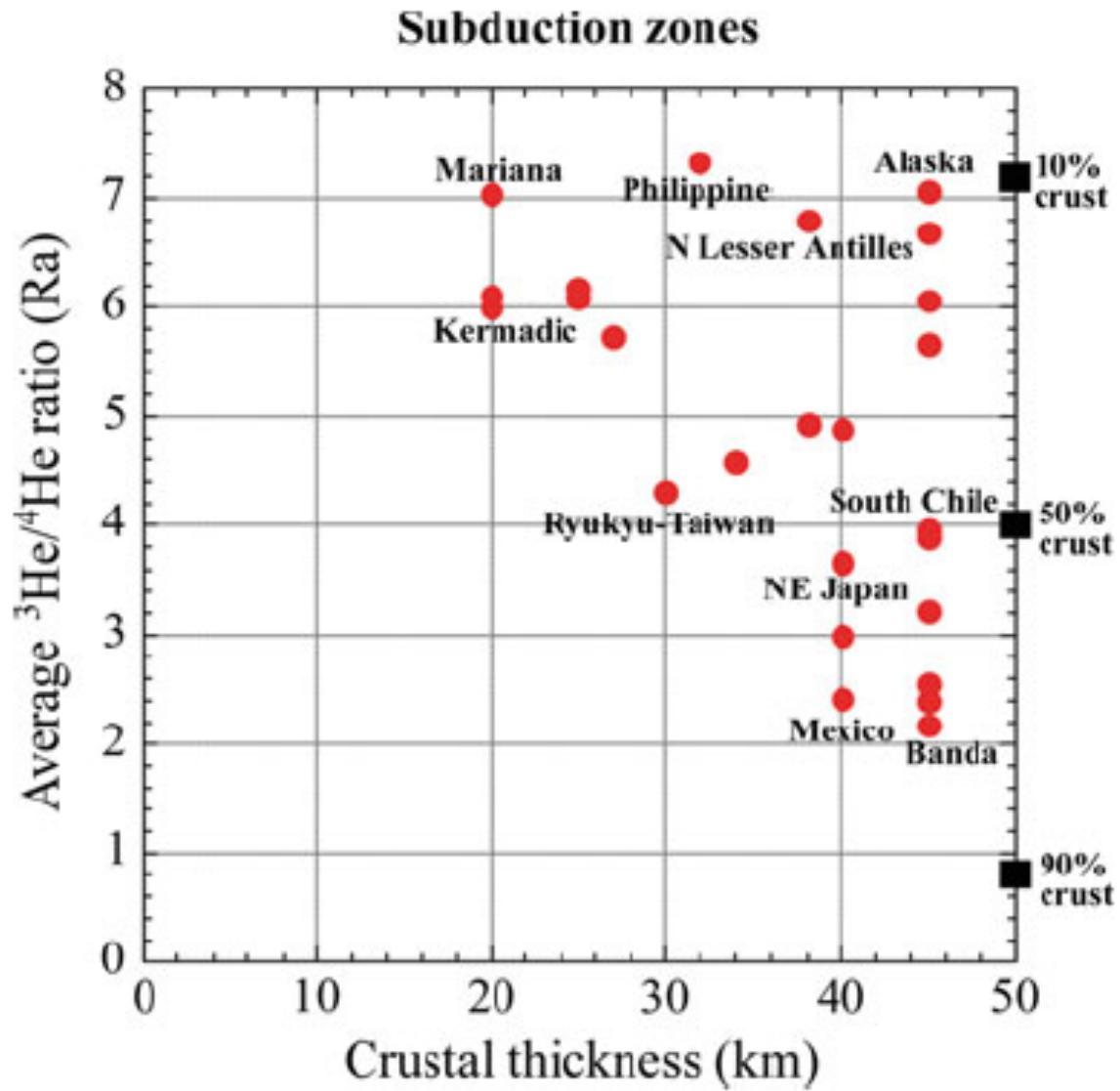


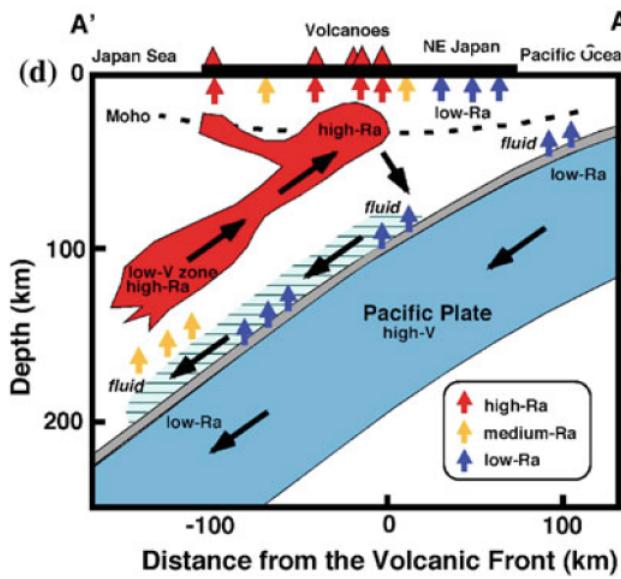
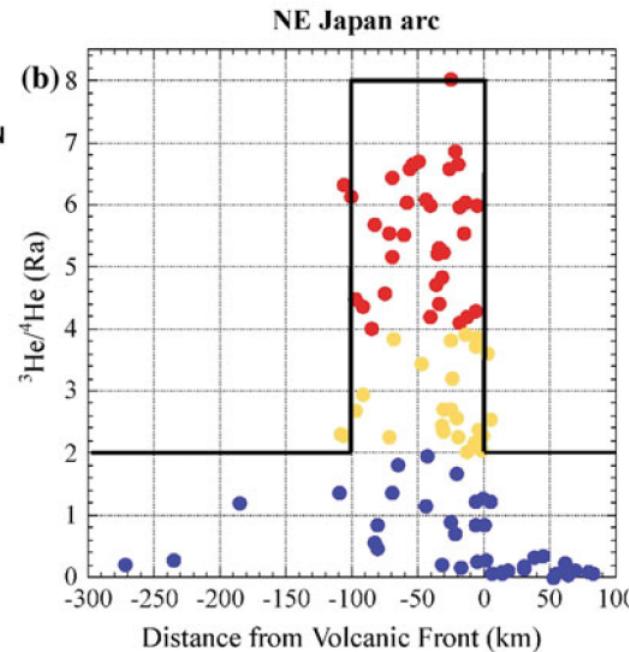
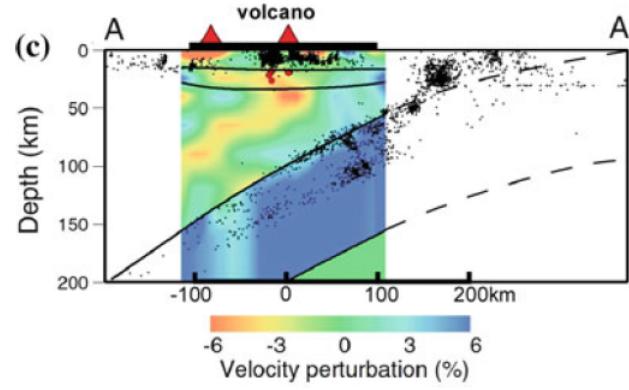
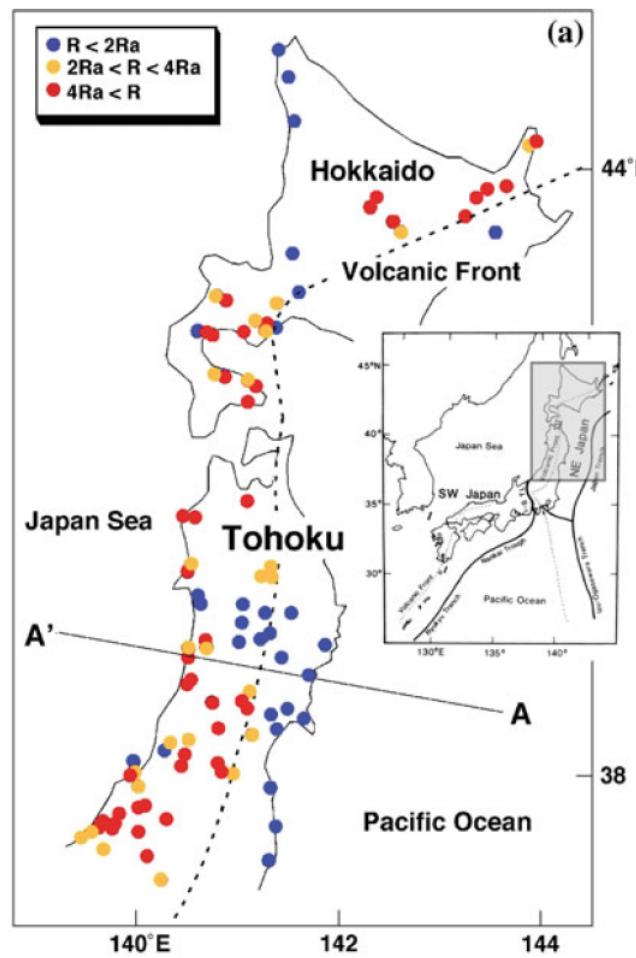
Sano and Fischer, 2013

Subduction Zones: Helium isotopes in fluids



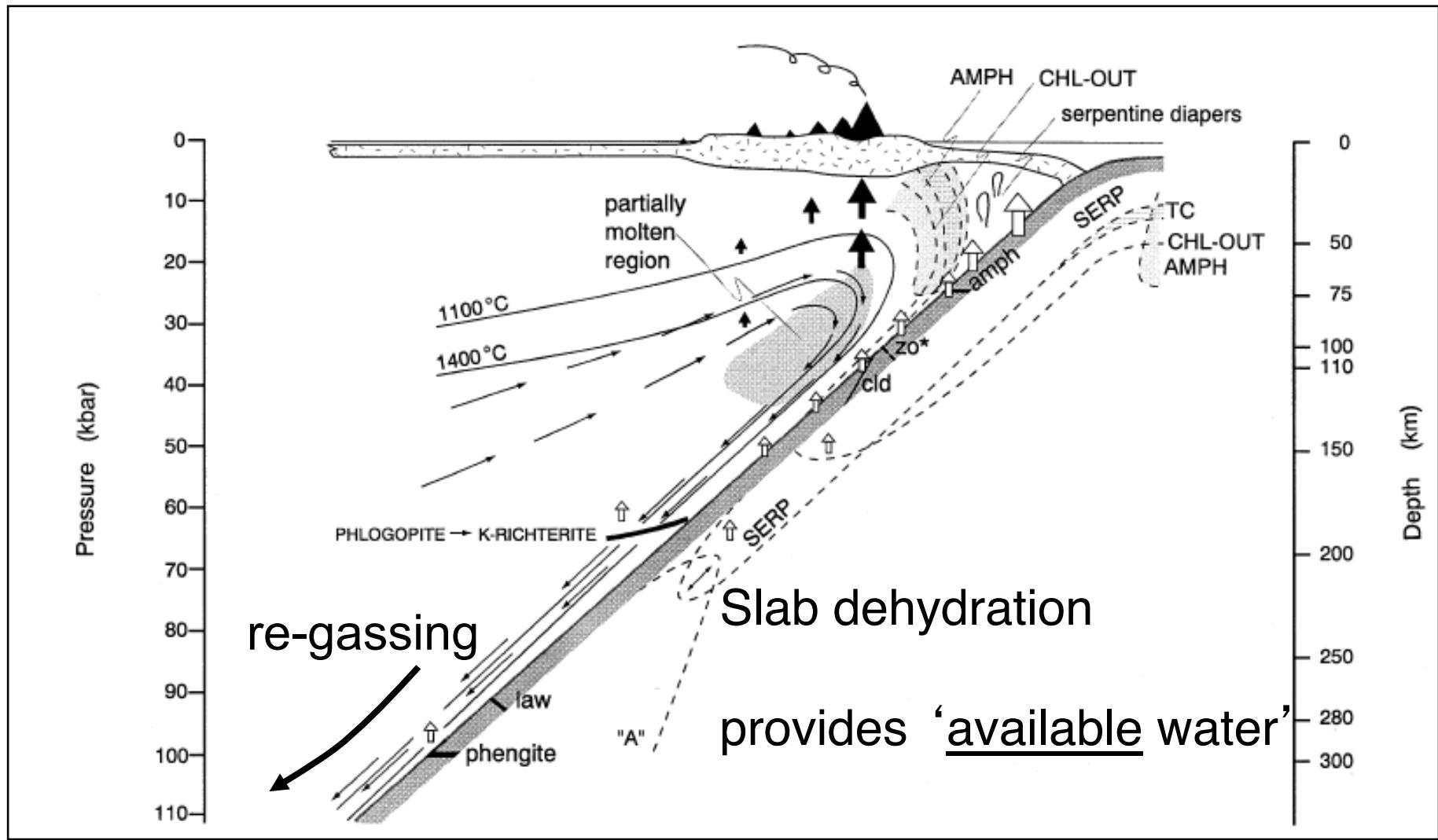
Sano and Fischer, 2013



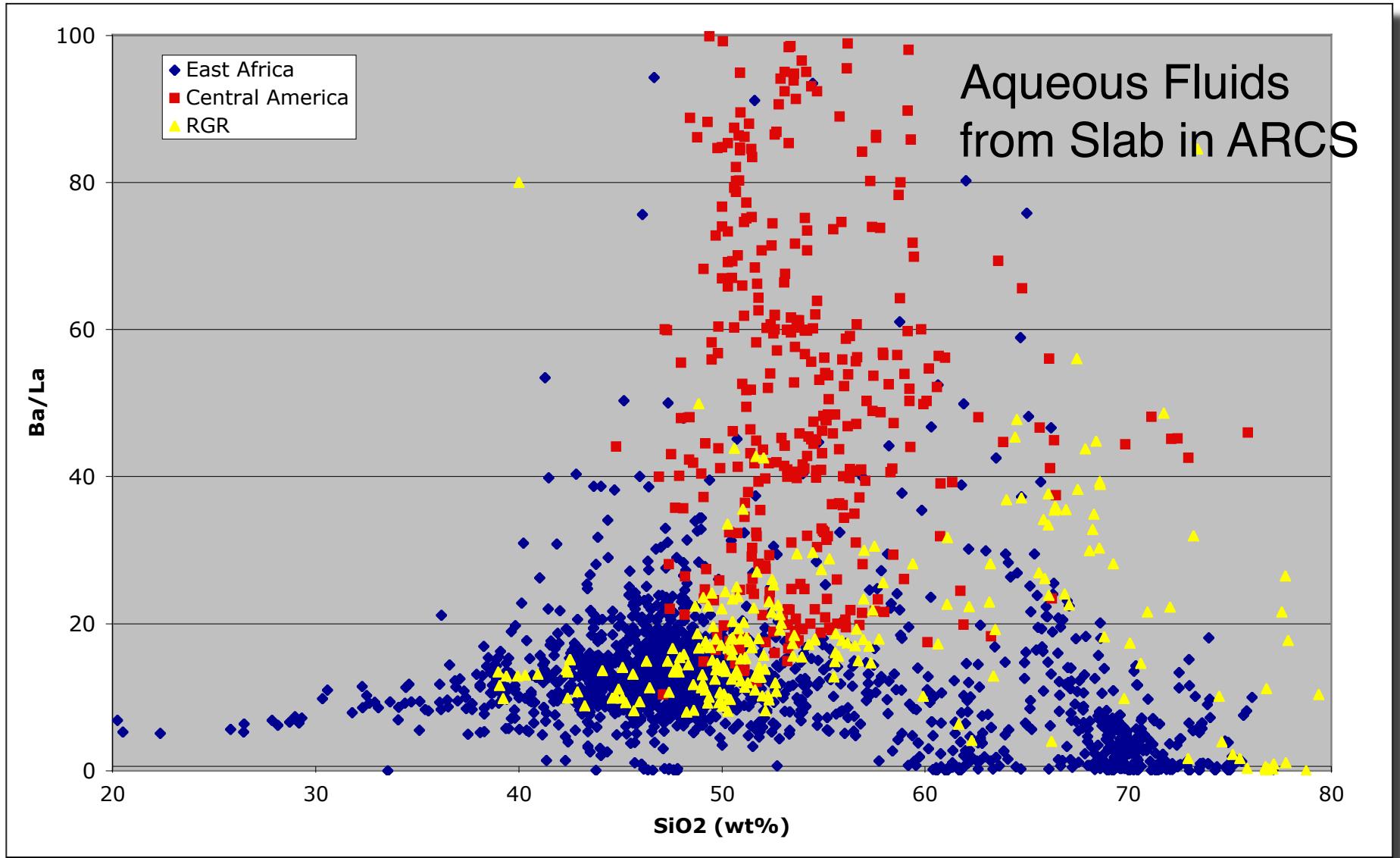


Sano and Fischer, 2013

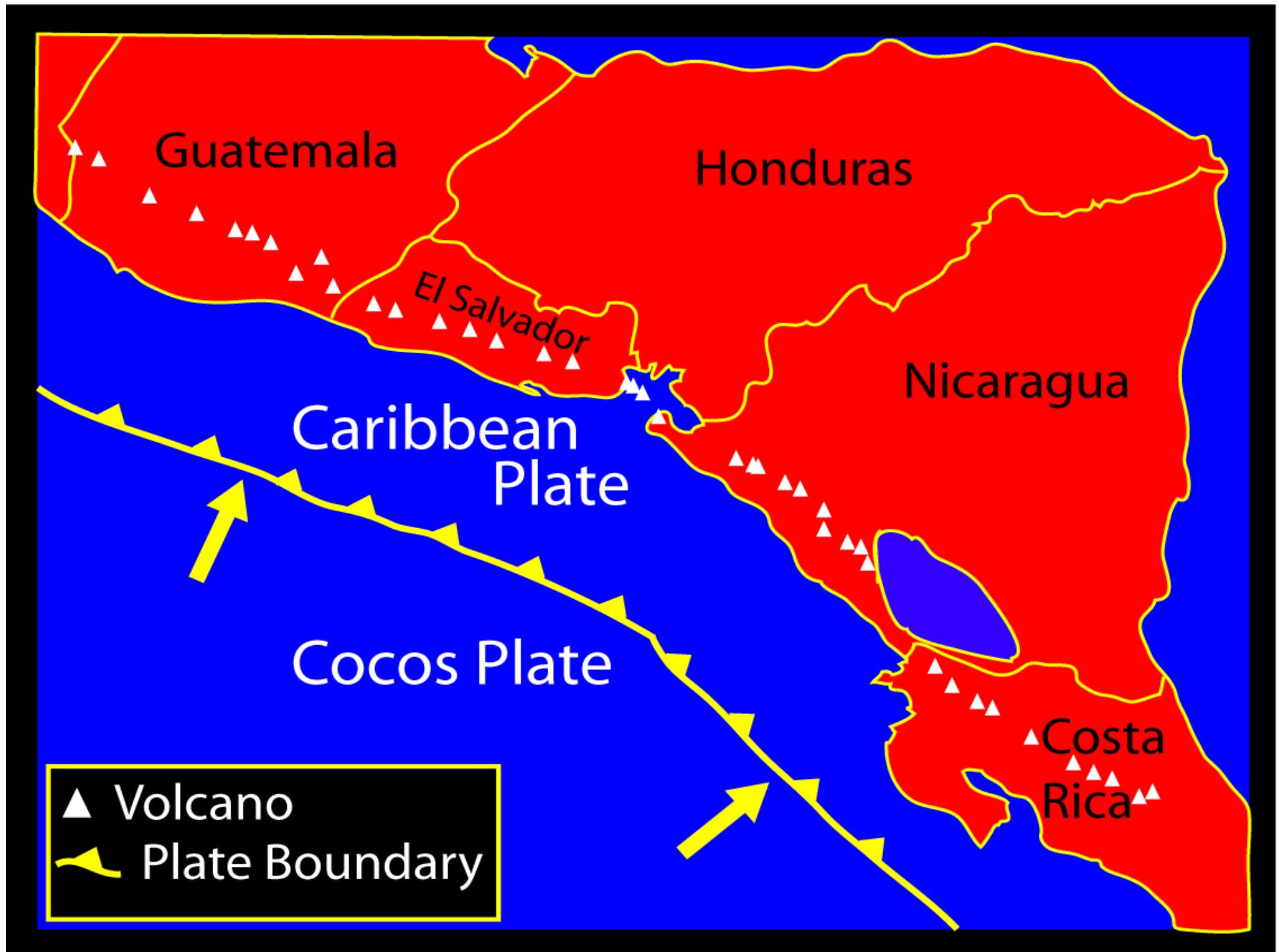
In Arcs fluids come from the slab



Schmidt+Poli, 1998



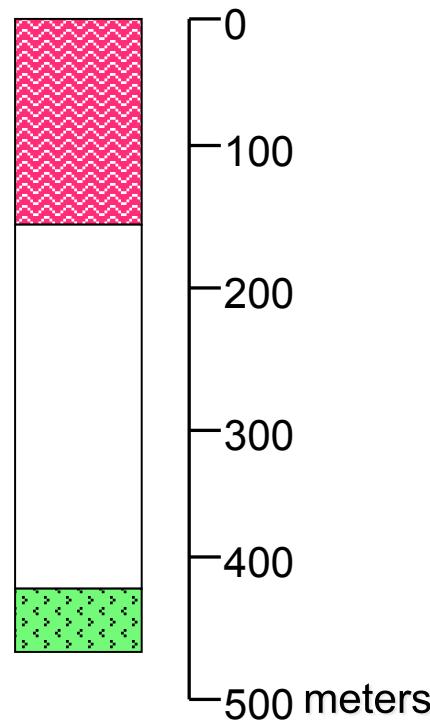
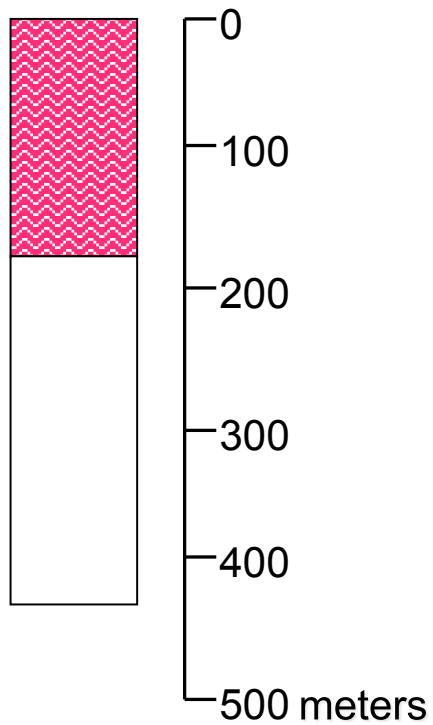
Georoc database and Mike Carr's data base



Sediment lithology:

★ Site 495
Guatemala

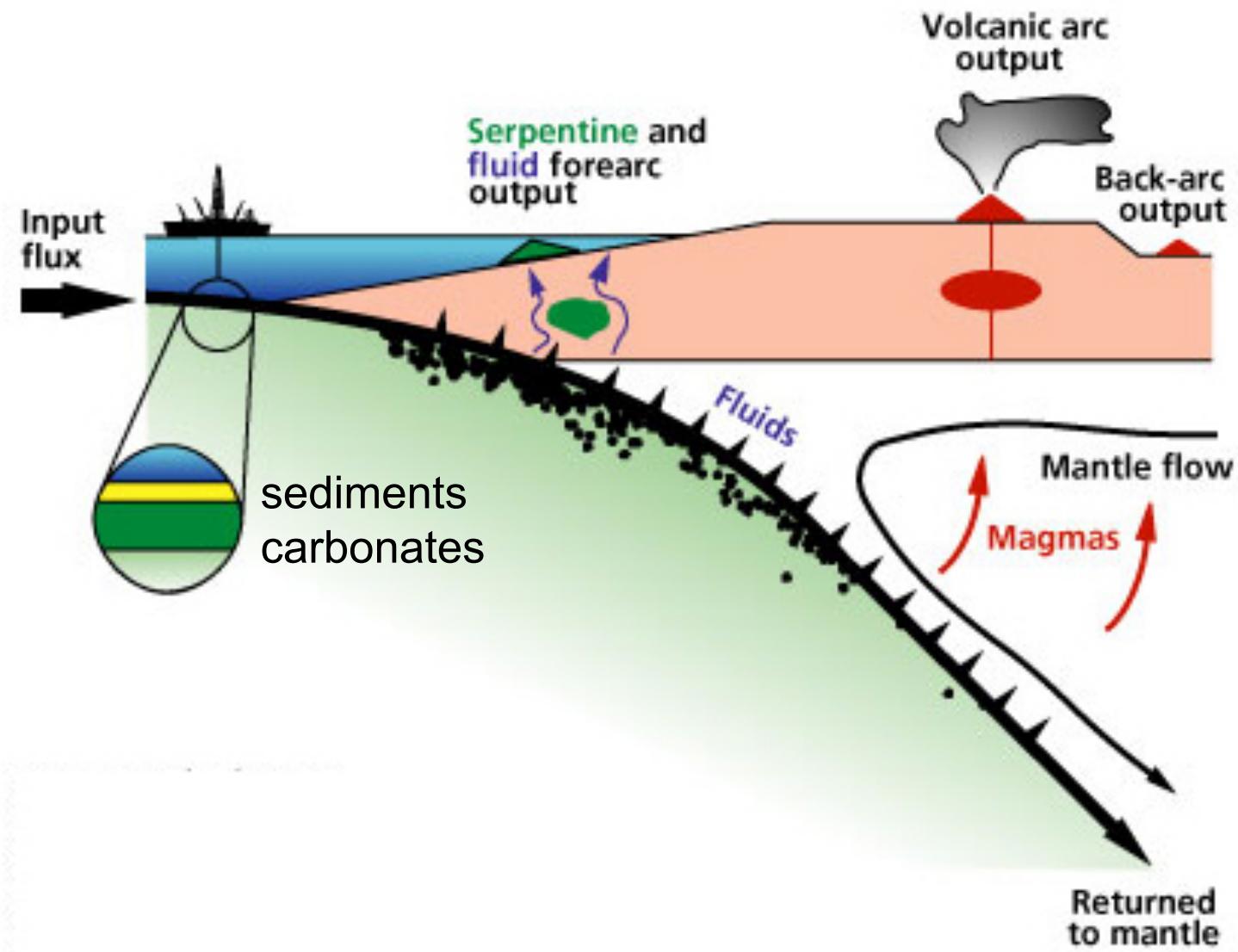
★ Site 1039
Costa Rica



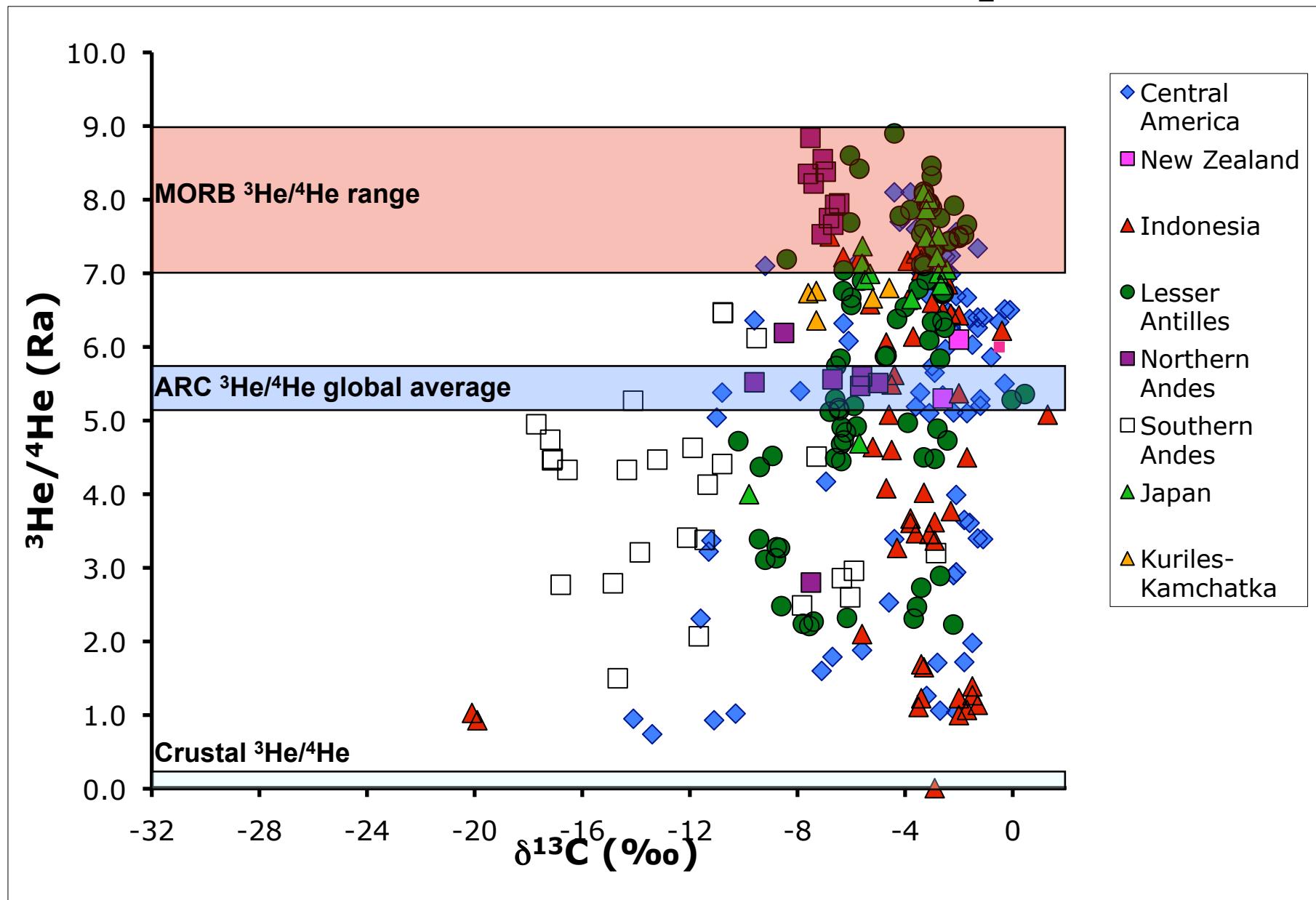
Hemi-pelagic
diatom-rich mud

Pelagic
Carbonates

Pyroxene Gabbro

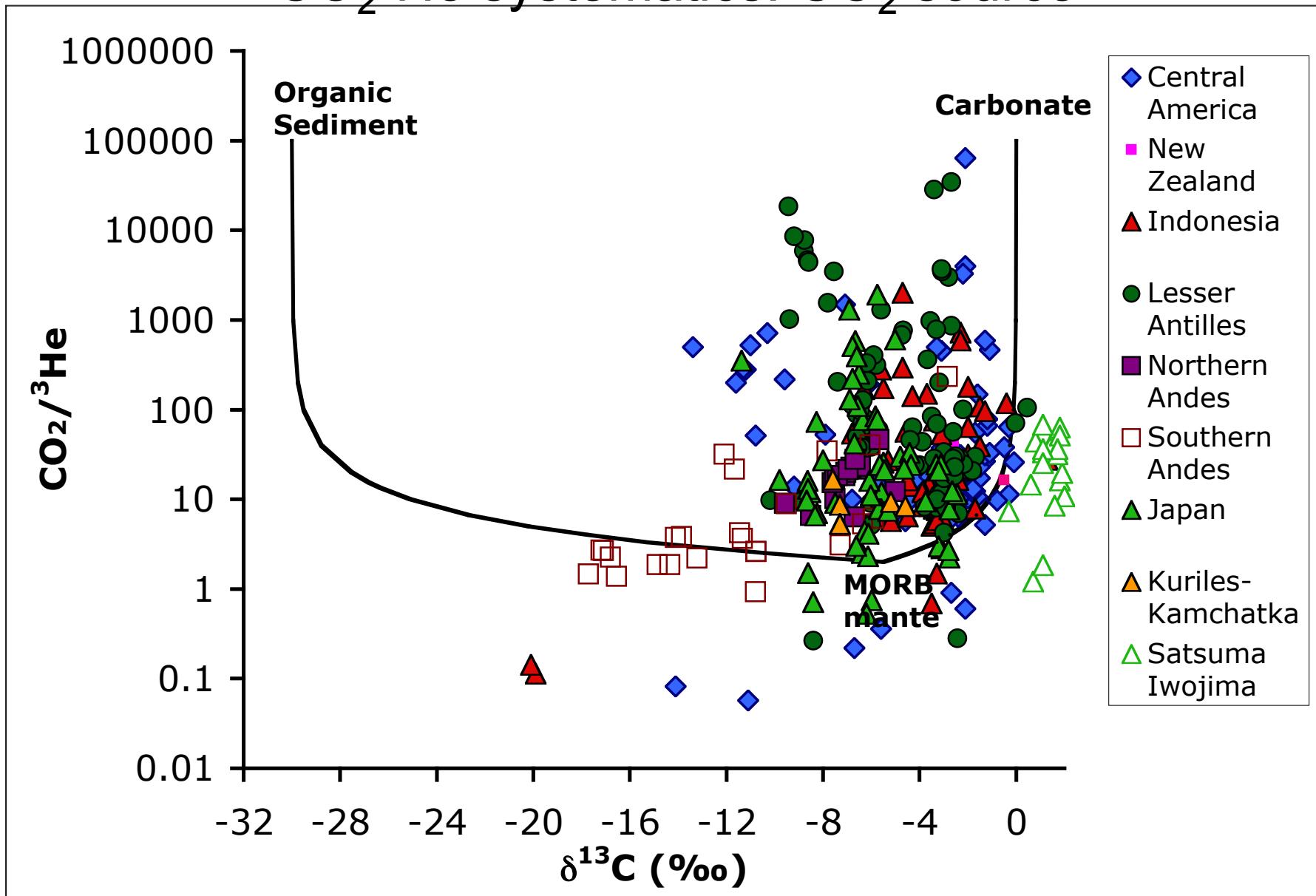


Arcs: Helium and Carbon isotopes (CO_2) in gases



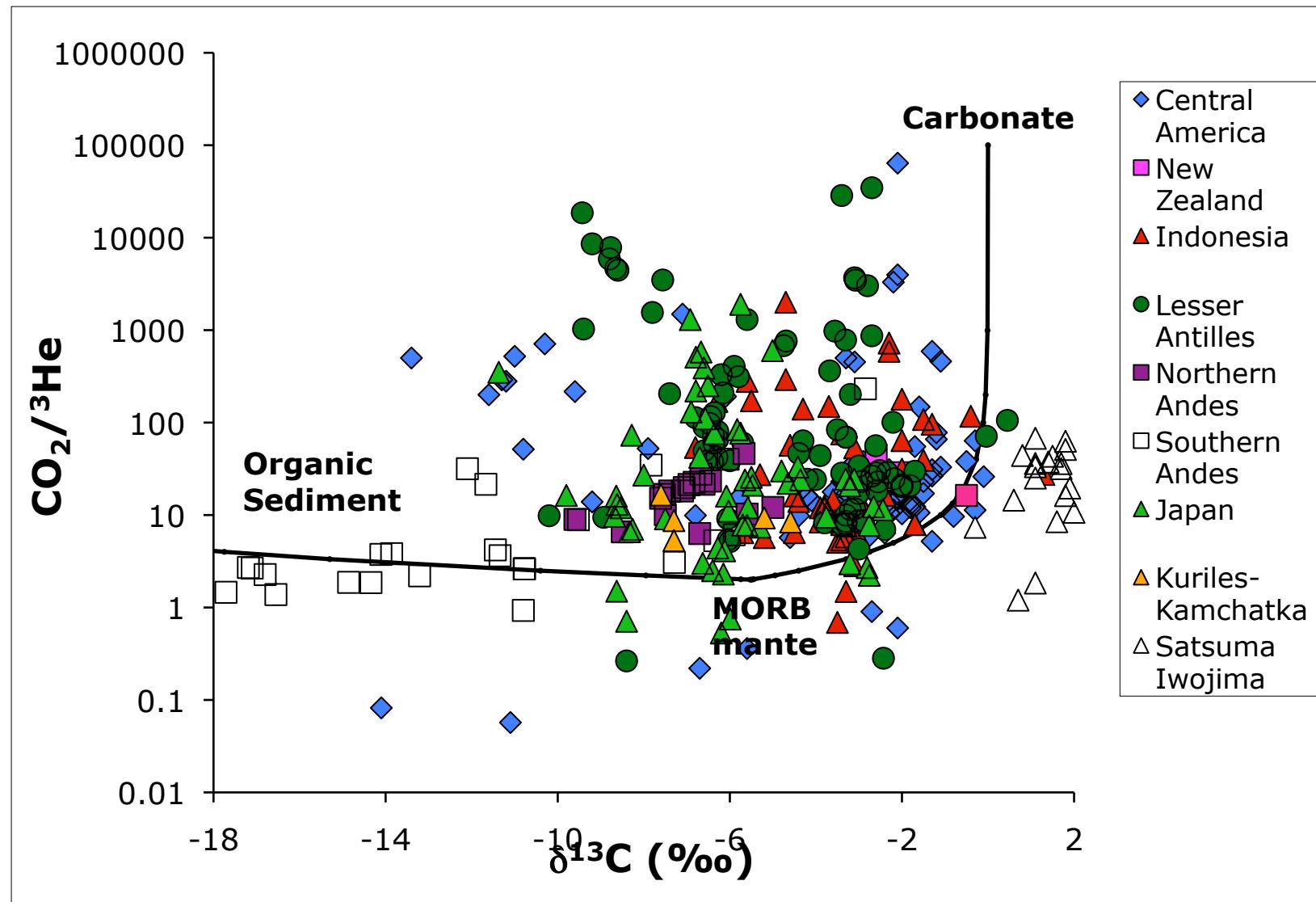
from Oppenheimer, Fischer, Scaillet (ToG in press)

CO_2 -He systematics: CO_2 source



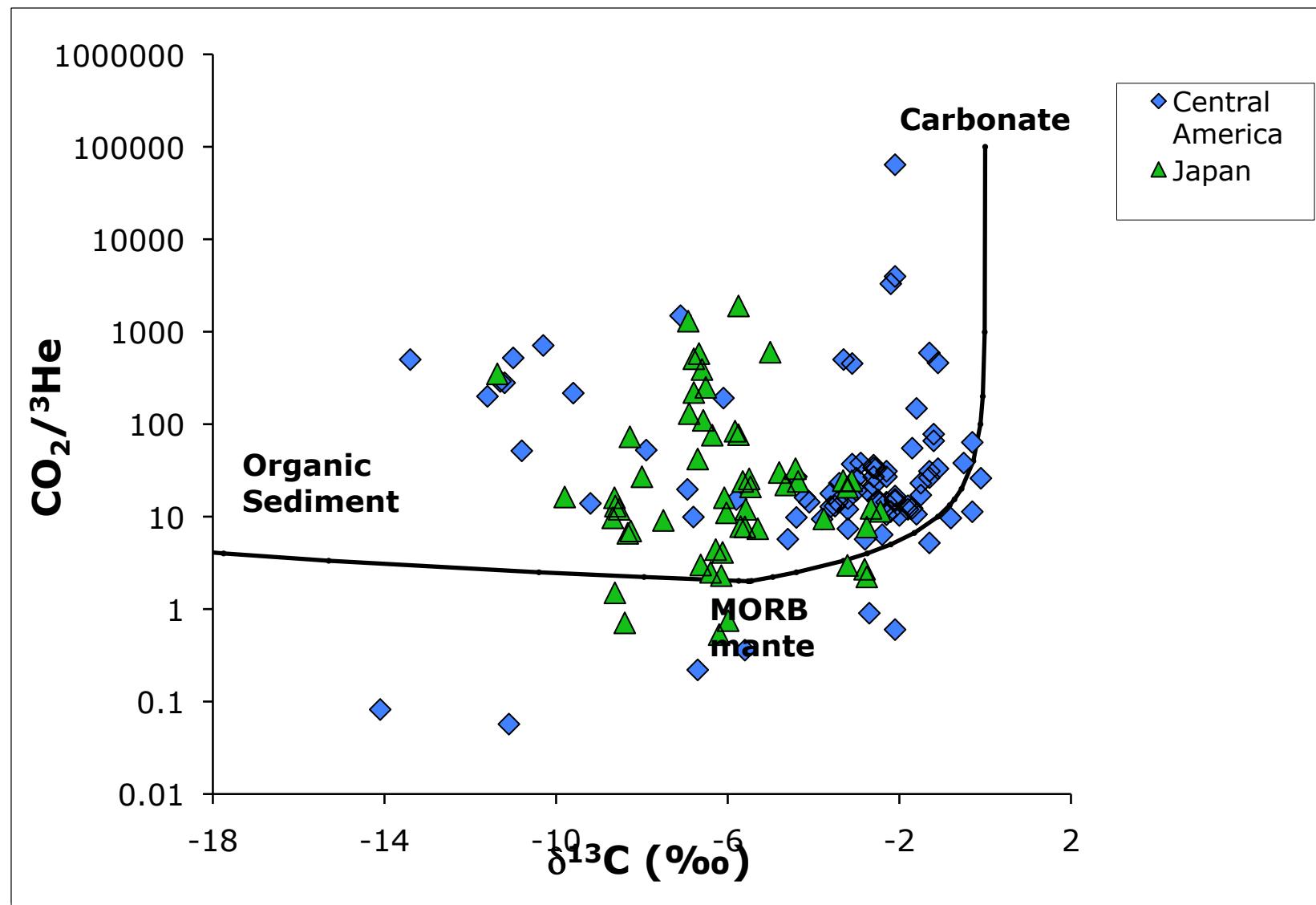
after Sano and Marty (1995); from Oppenheimer, Fischer, Scaillet (ToG in press)

CO_2 -He systematics: CO_2 source



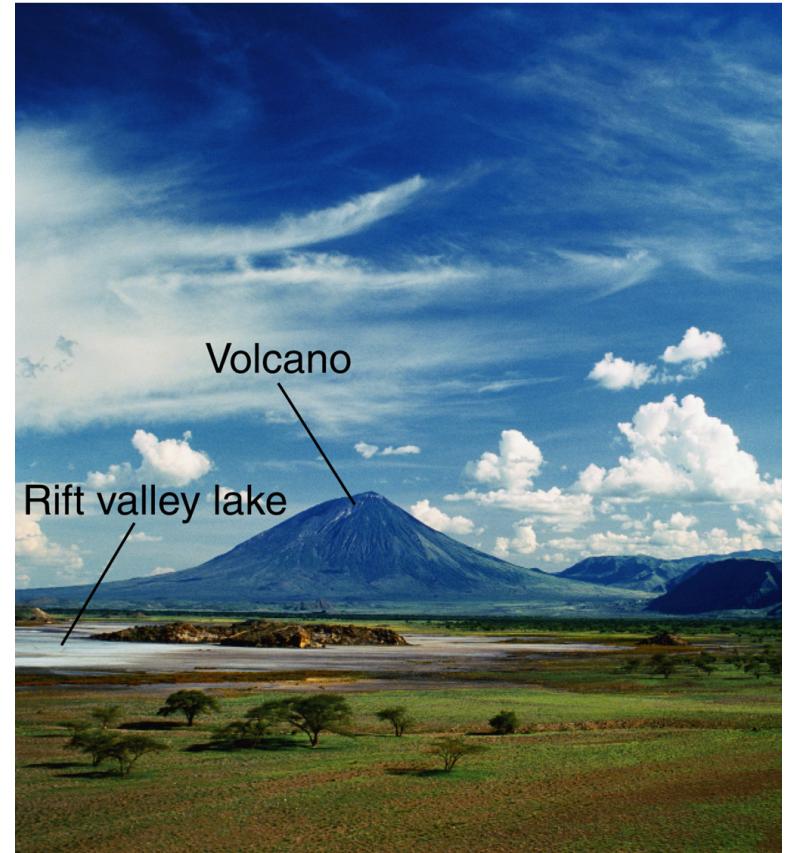
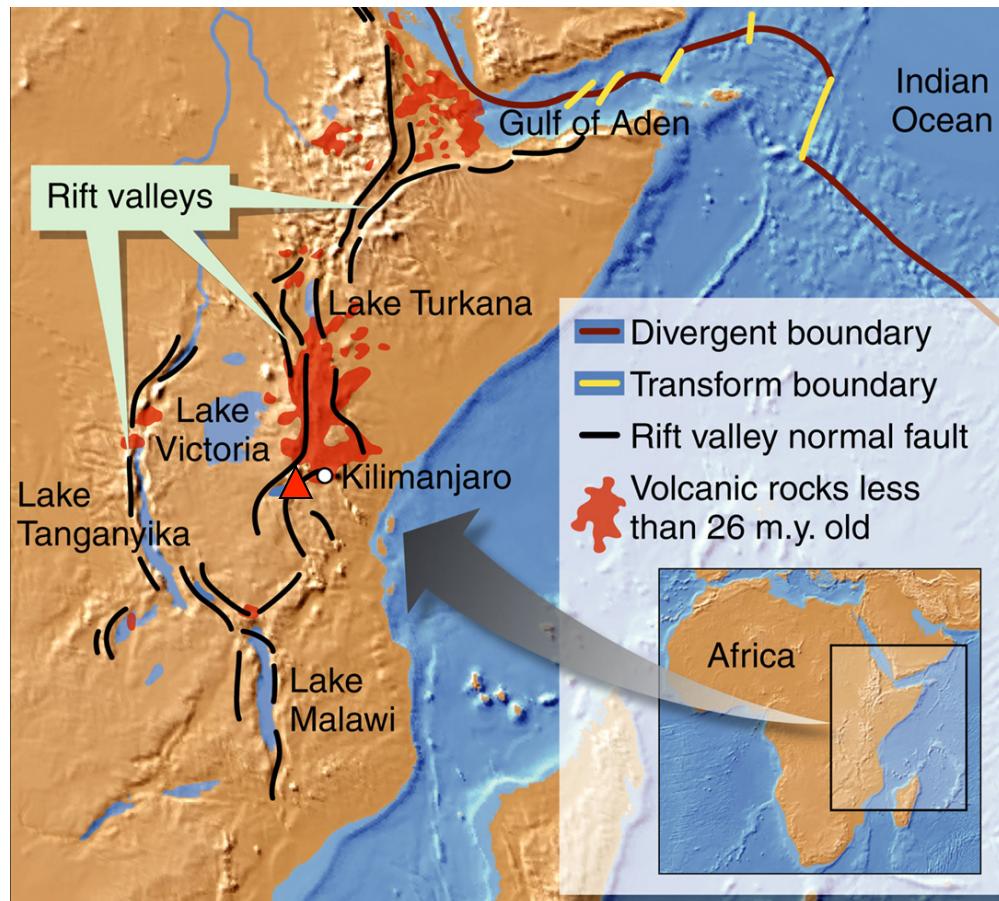
after Sano and Marty (1995); from Oppenheimer, Fischer, Scaillet (ToG in press)

CO_2 -He systematics: CO_2 source



after Sano and Marty (1995); from Oppenheimer, Fischer, Scaillet (ToG in press)

Upper mantle gases away from Subduction Zones: Oldoinyo Lengai



Oldoinyo Lengai:
World's only active carbonatite
Volcano

From Smith & Pun "How does Earth Work?"

OI Doinyo Lengai Gases:

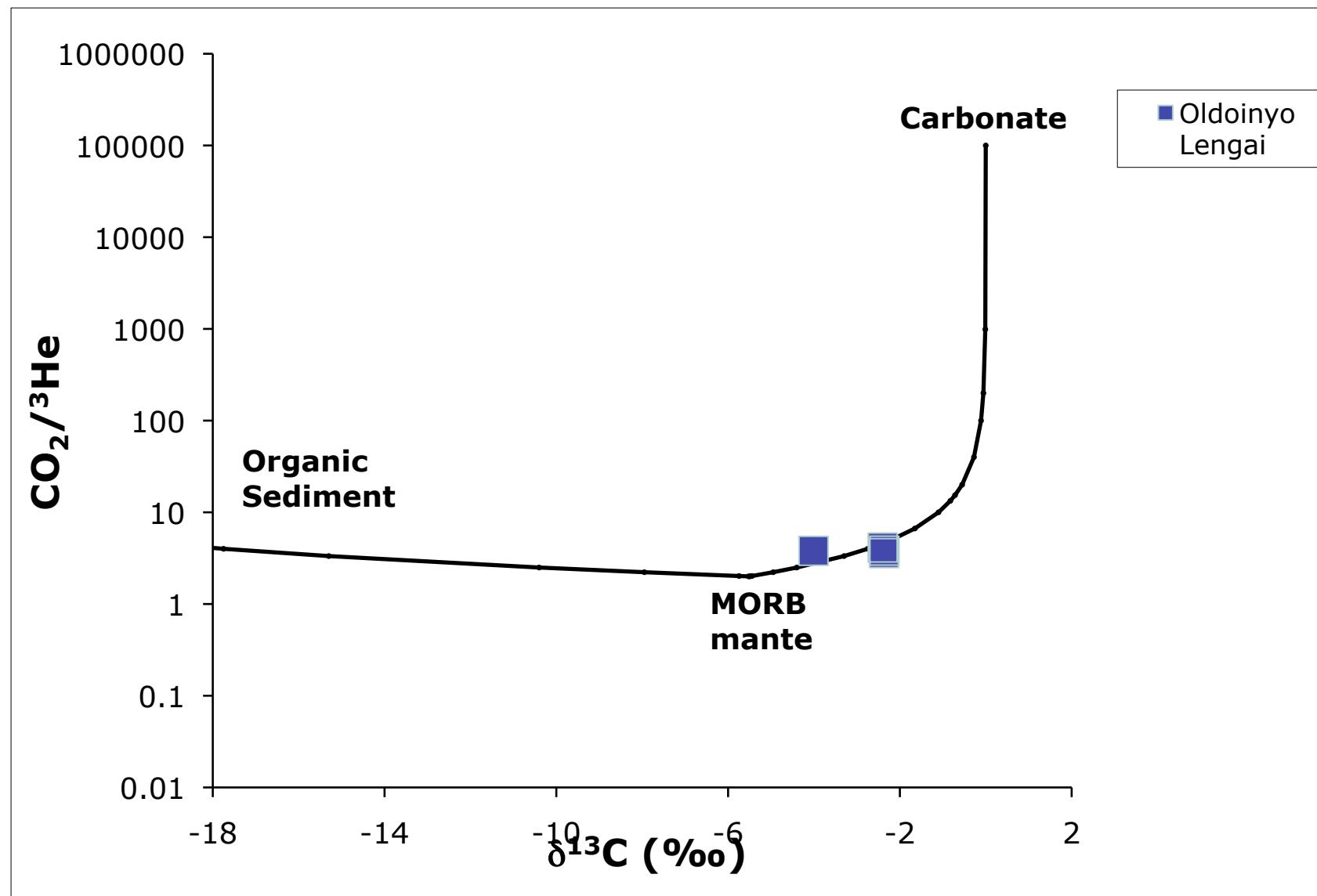
${}^3\text{He}/{}^4\text{He}$: $6.86 \pm 0.03 \text{ R}_\text{A}$

(Fischer et al., 2009)



July 2005

CO_2 -He systematics: CO_2 source

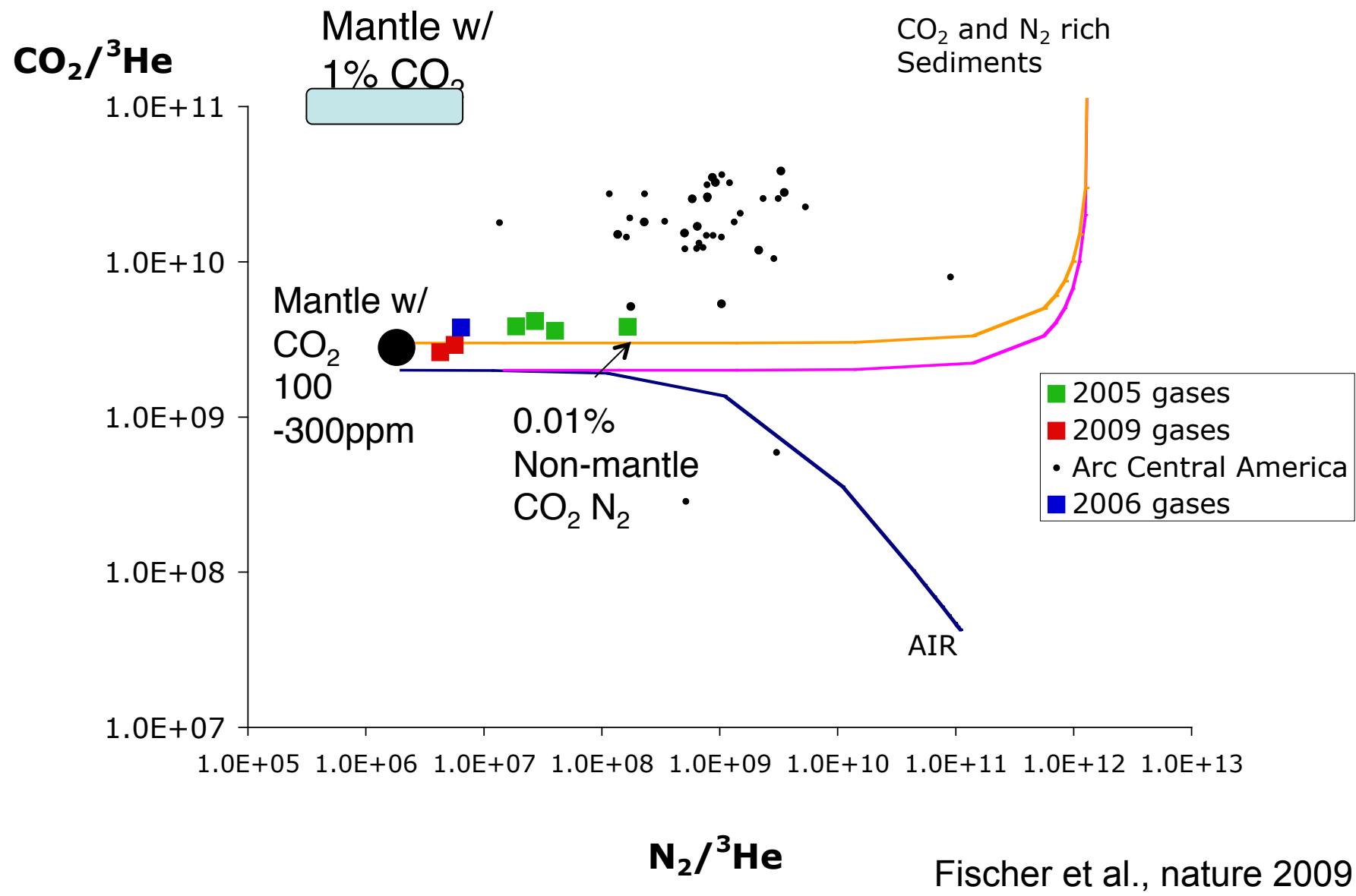


after Sano and Marty (1995); Fischer et al., nature 2009

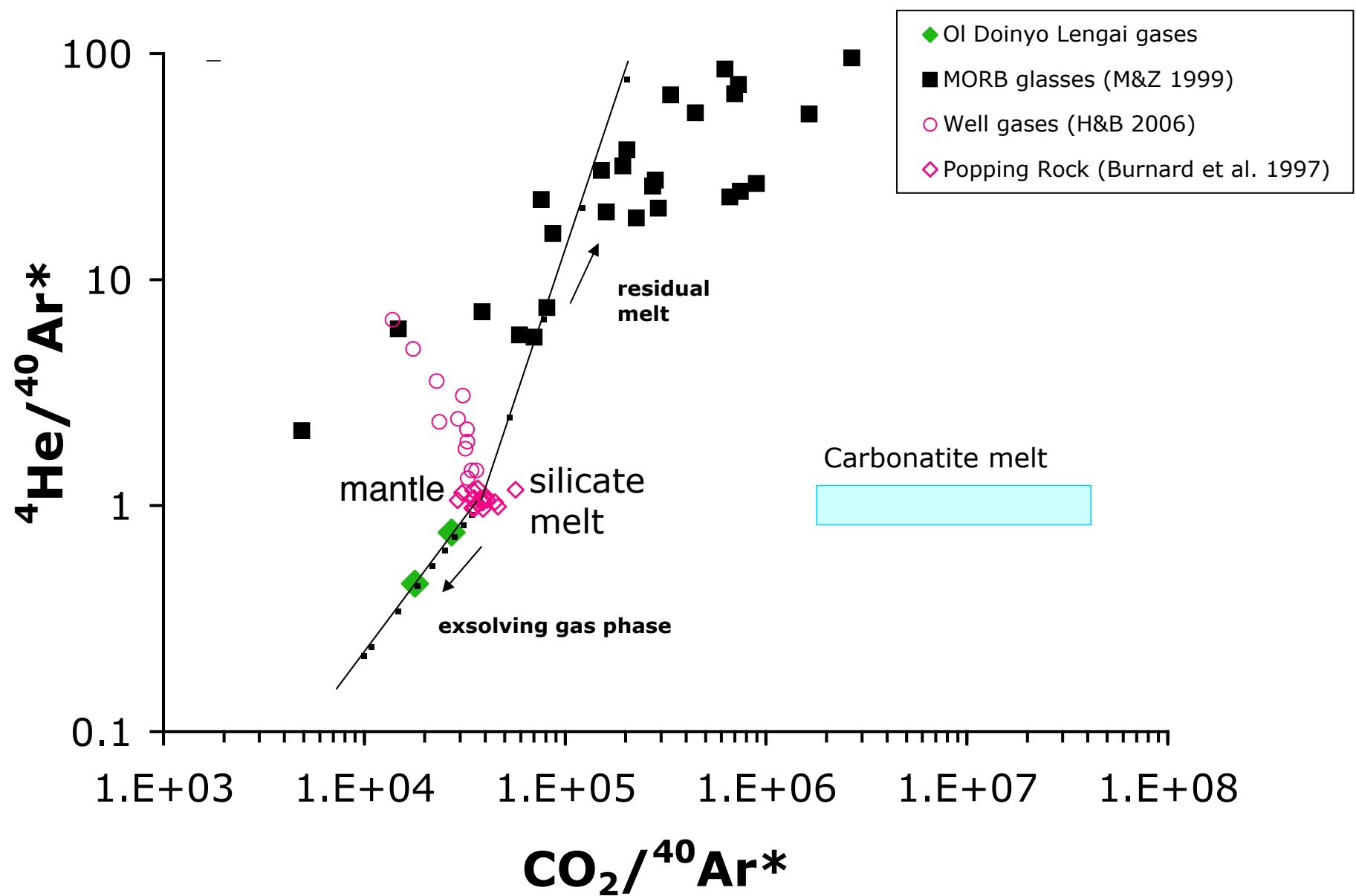
CO_2 , Nitrogen, Chlorine in Magma Source

Year	${}^3\text{He}/{}^4\text{He} R_c / R_A$	$\text{CO}_2/{}^3\text{He}$ $(\times 10^9)$	$\delta^{13}\text{C}$ $(\text{\textperthousand})$	$\delta^{15}\text{N}$ $(\text{\textperthousand})$	$\delta^{37}\text{Cl}$ $(\text{\textperthousand})$
2005	6.77 ± 0.16	2.60	-2.43 ± 0.02	-3.5 ± 0.4	0.0 ± 0.3
2009	6.88 ± 0.20	2.91	-2.77 ± 0.04	-1.5 ± 0.4	
MORB	8 ± 1	2.5 to 3.0	-5 to -2.5	-5 ± 3	0 ± 0.5

CO_2 in Magma Source Region

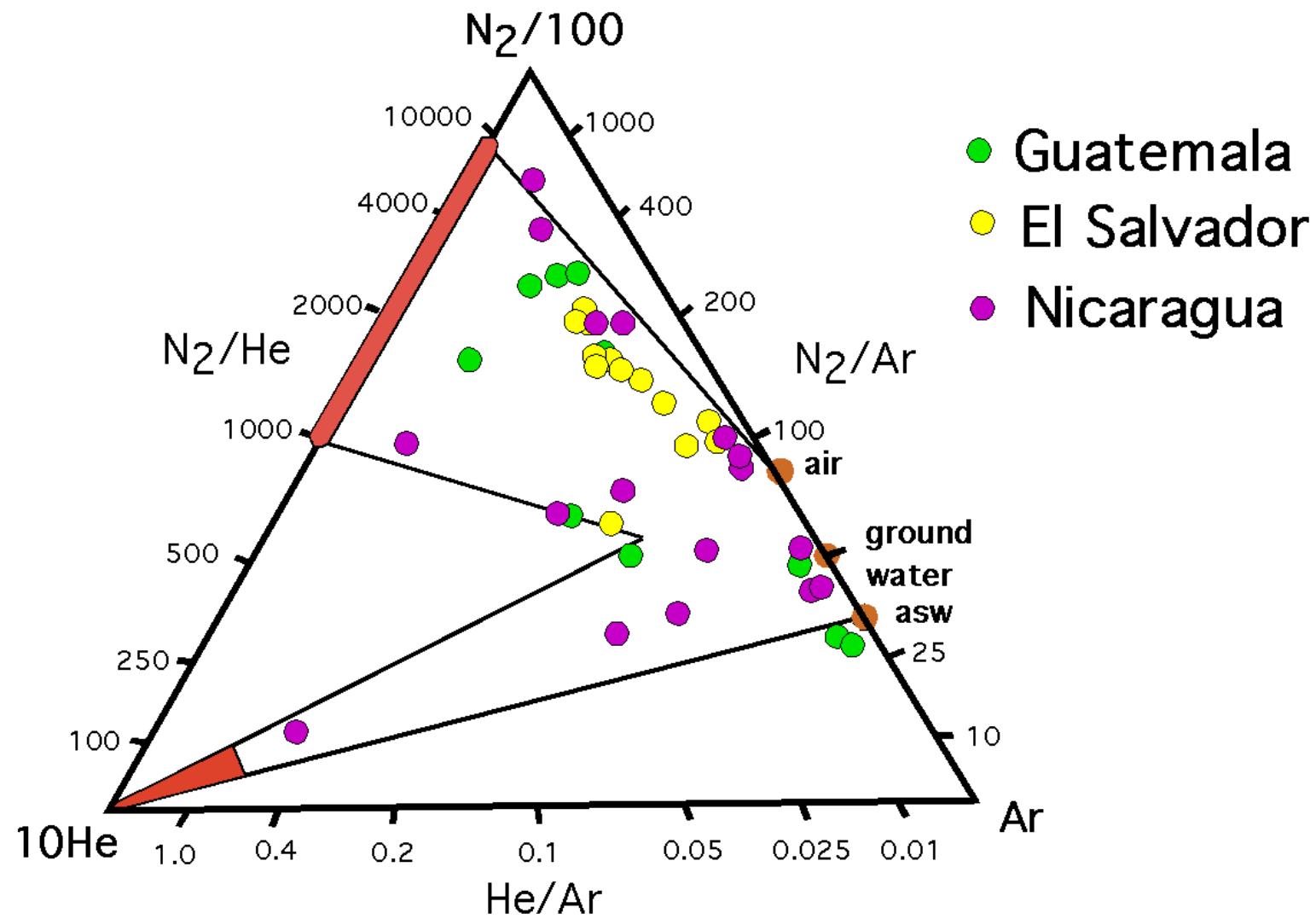


CO_2 -He-Ar systematics: mantle-derived silicate melt degassing

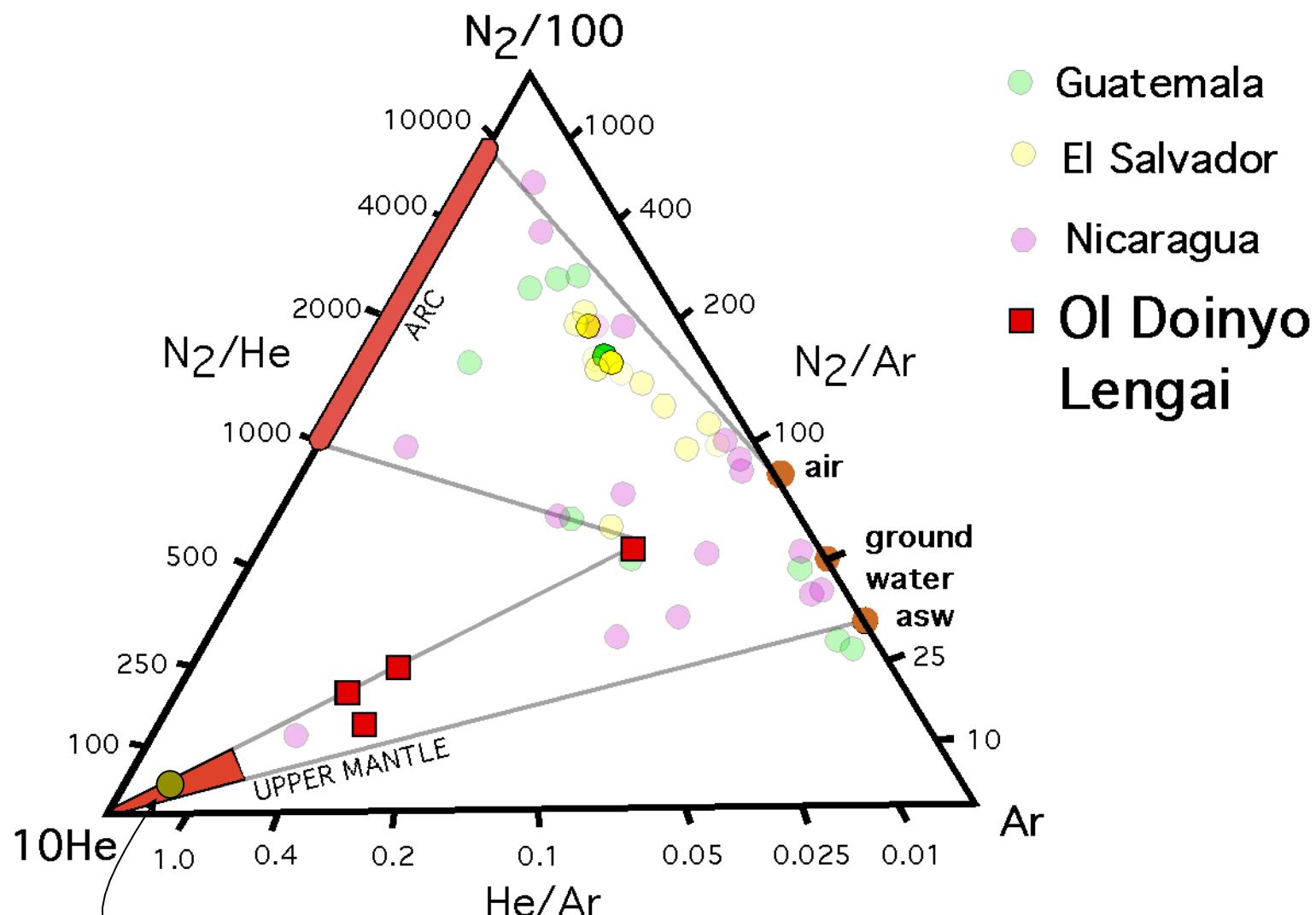


Fischer et al., nature 2009

Gas Samples



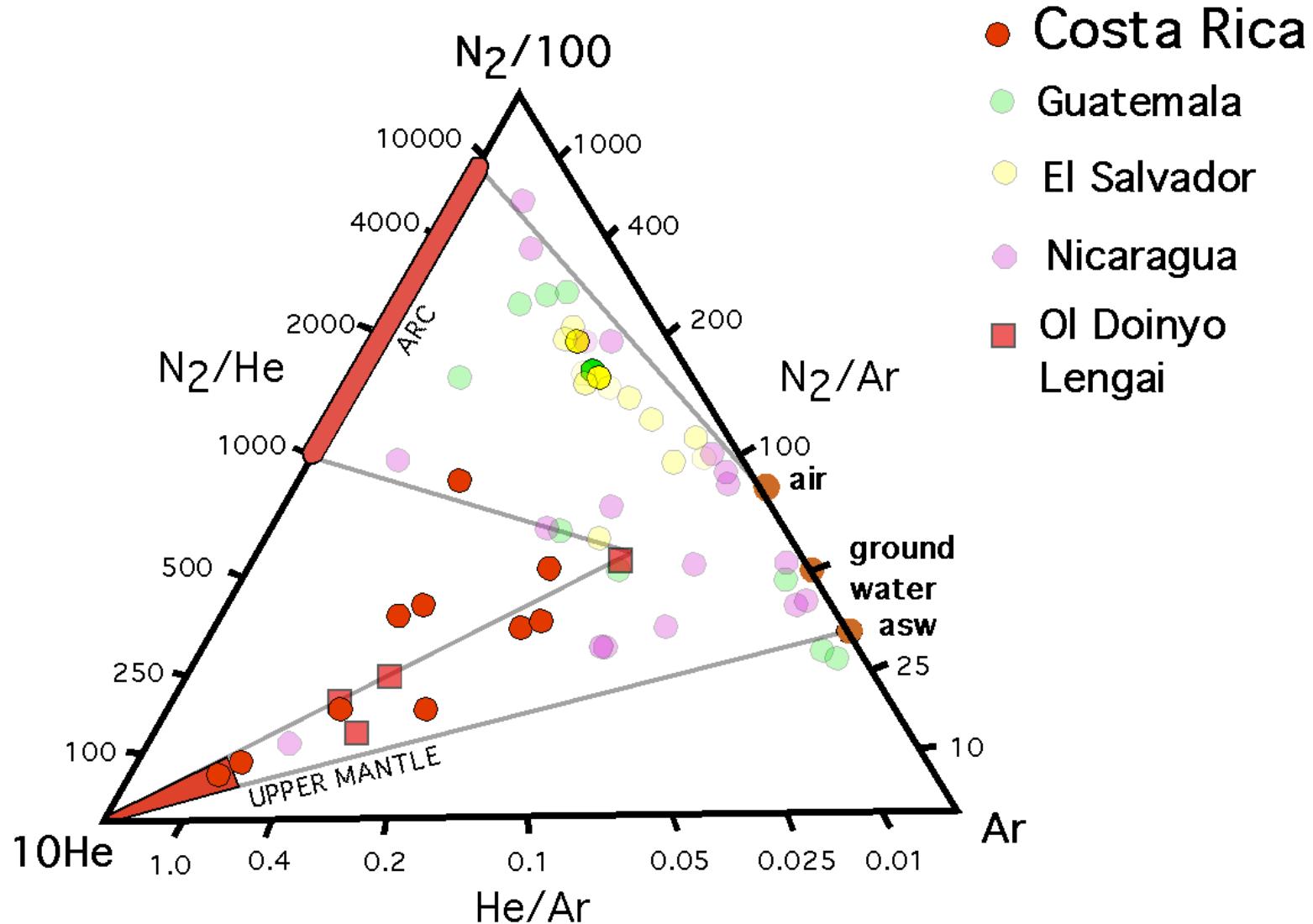
Zimmer, Fischer, et al. G3, 2004; Fischer unpubl.



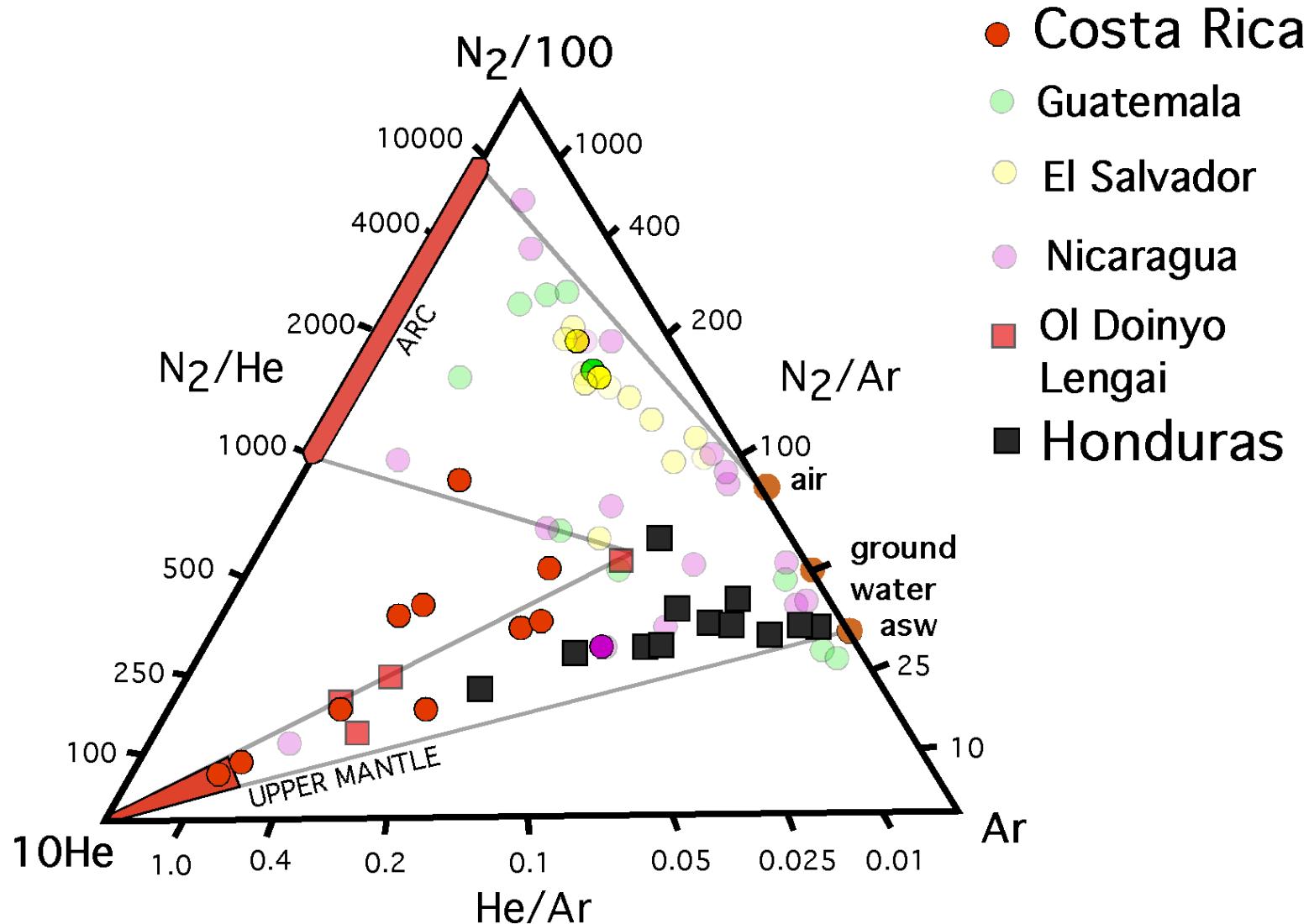
- Guatemala
- El Salvador
- Nicaragua
- Ol Doinyo
Lengai

Mid-Atlantic Popping Rocks
(Javoy & Pineau, '91)

Zimmer, Fischer, et al. G3, 2004; Fischer unpubl.



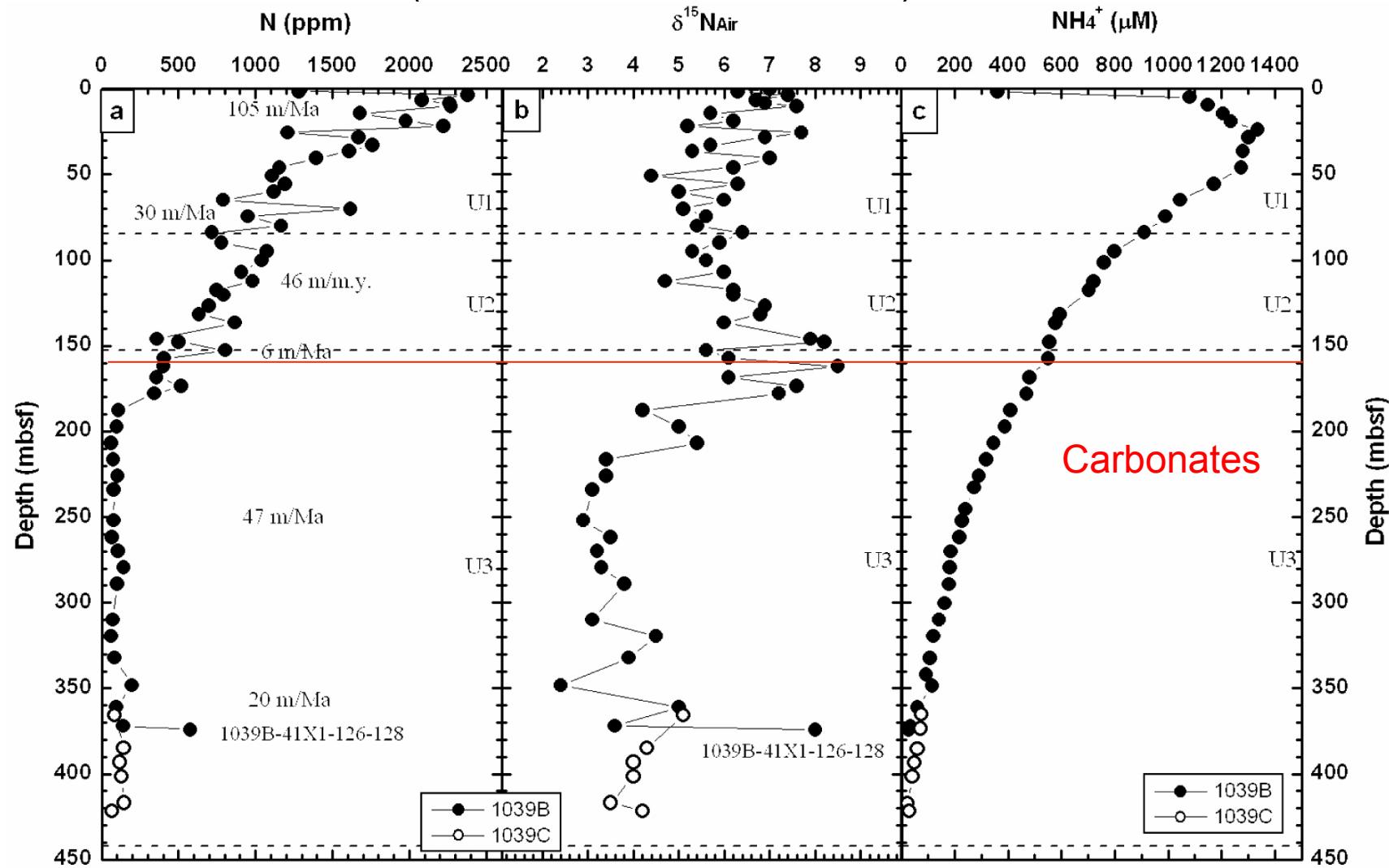
Zimmer, Fischer, et al. G3, 2004; Fischer unpubl.
Fischer et al., nature 2009

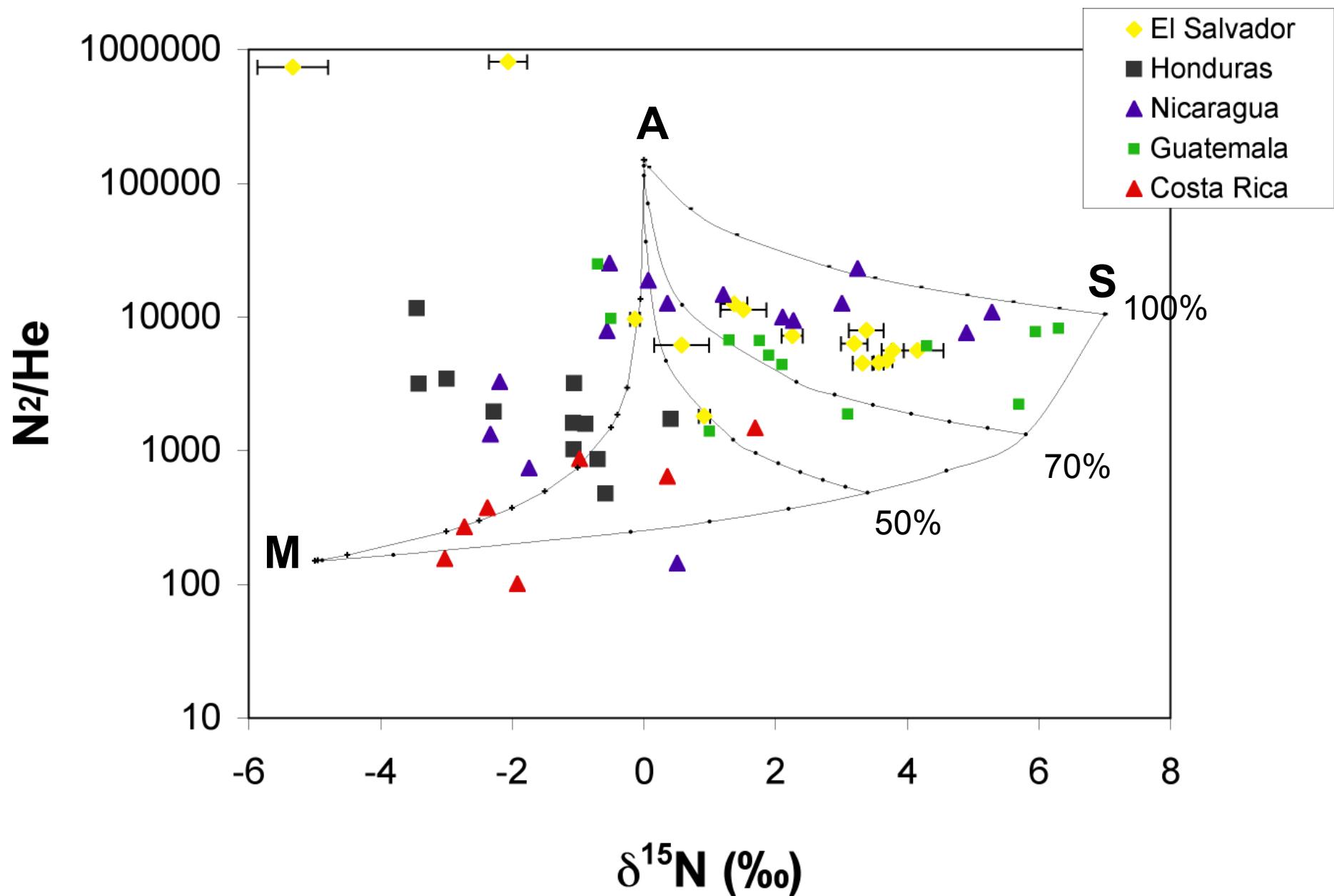


Zimmer, Fischer, et al. G3, 2004; Fischer unpubl.
Fischer et al., nature 2009

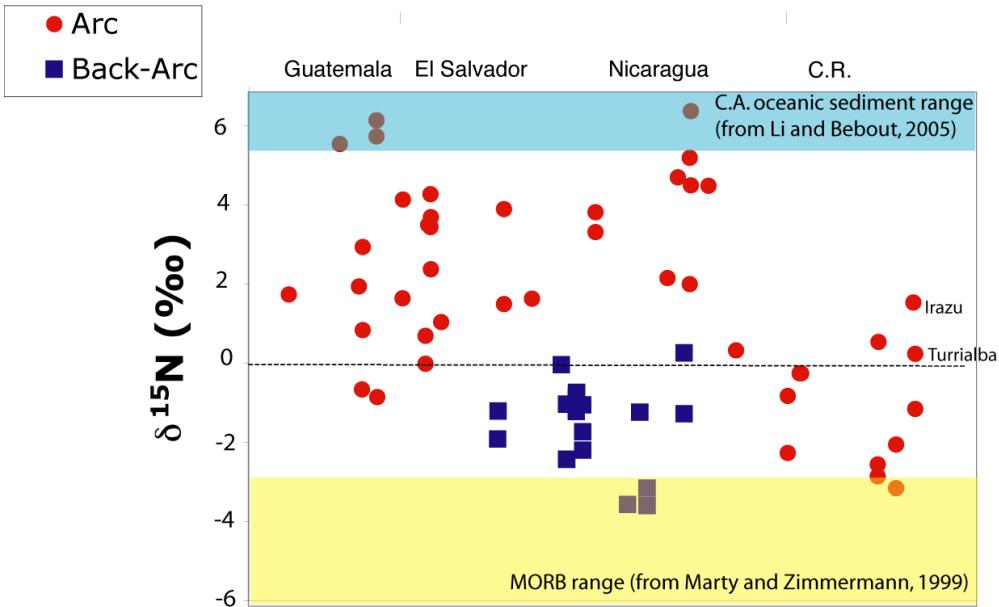
Nitrogen in Sediments of Site 1039: Off Costa Rica

(Li and Bebout, JGR 2005)



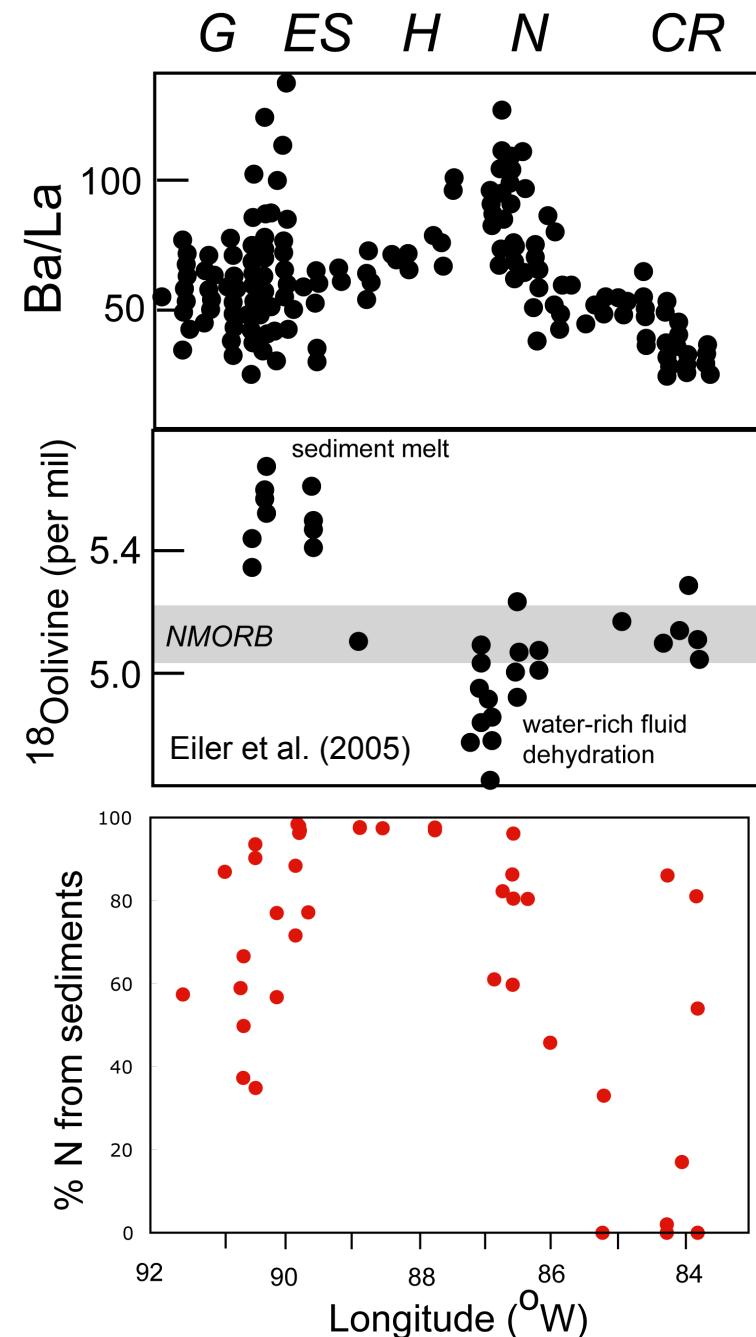


Fischer et al., Science, 2002; Zimmer, Fischer, et al. G3, 2004; Elkins, Fischer et al., GCA 2006



Use end-members
to calculate

Elkins et al. (2006), Fischer et al. (2002),
Zimmer et al. (2004), M. Carr's online database



Suggestions for beginning gas/ geochemical monitoring and research

- Start technically simple: alkaline solutions, Gigganbach bottle, multiGAS
- Establish baselines of S/Cl and S/CO₂
 - > do your own analyses
- Collaborate for noble gas and stable isotopes
 - > technology transfer
- Build chemistry lab in central place
- Work up to other techniques: Picarro C, O, H